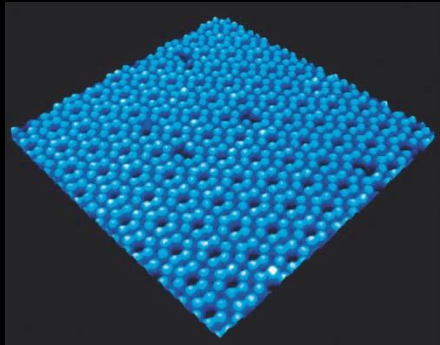


Earth Materials

Minerals & Rocks



Matter

An atom is the smallest unit of an element that possesses the properties of the element. It consists of a nucleus of protons and neutrons and a surrounding cloud of electrons. There are three states of matter: gas, liquid, and solid. Each state is distinguished by unique physical properties. Processes in Earth's dynamics mostly involve the changing of matter from one state to another.



Nature of Minerals

A mineral is a natural inorganic solid with a specific internal structure and a chemical composition that varies only within specific limits. All specimens of a given mineral, regardless of where, when, or how they were formed, have the same physical properties (including cleavage, crystal form, hardness, density, color, luster, and streak). Minerals also have restricted stability ranges.



Earth Materials

Minerals & Rocks

- Mineral adalah **benda padat homogen yang terbentuk dalam, inorganik dan mempunyai struktur kristal padat, dengan komposisi kimia tertentu dan mempunyai sifat fisik tertentu**
- Setiap mineral mempunyai susunan atom yang teratur membentuk kristal padat.
- *Native Elements* merupakan mineral yg tidak umum, tetapi beberapa diantaranya mempunyai nilai ekonomik yang tinggi seperti emas, tembaga, perak, diamond dan graphite.
- Mineral dapat mempunyai komposisi kimia yang bervariasi. Variasi ini terjadi akibat proses substitusi atom dari satu elemen ke elemen lainnya. Substitusi ini dapat terjadi apabila kesetimbangan elektrik terjadi dalam struktur atom, dan jika elemen yang mensubstitusi mempunyai jari-2 atom yang hampir sama.
- Lebih dari 3500 mineral adalah silikat (gabungan Si, O dan elemen lainnya). Seperti Ferromagnesian silikat mengandung Fe dan Mg. Group mineral lainnya adalah karbonat, sulfida, sulfat, dan halida.



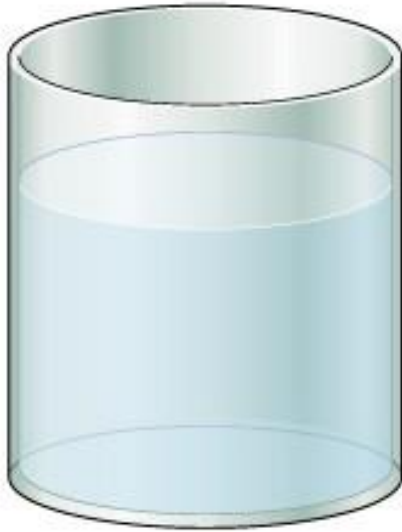
Earth Materials

Atom & Element

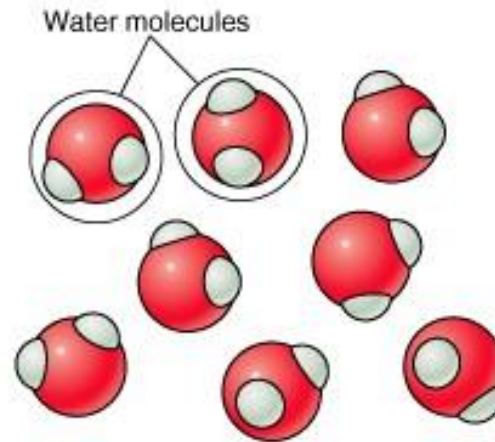
- Setiap materi terdiri dari atom.
- Atom terdiri dari inti (*nucleus*) yang terdiri dari *proton* dan *neutron* serta *electron* yang mengelilinginya.
- Elemen didefinisikan berdasarkan jumlah *proton* didalam inti.
- Masa Atom (atomic mass) adalah jumlah proton dan neutron dalam inti.
- Elemen dengan bentuk yang sama tetapi mempunyai masa berbeda dinamakan isotop.
- Jumlah elektron yang mengelilingi inti sama dengan jumlah proton didalam inti.
- Atom saling bergabung dikarenakan gaya ikat (bonding).
- Atom dari elemen-2 yang berbeda saling terikat membentuk *compounds*.
- Pada umumnya mineral adalah *compound*, tetapi hanya sedikit yang mempunyai satu jenis elemen yang dikenal sebagai *native elements*.

Earth Materials

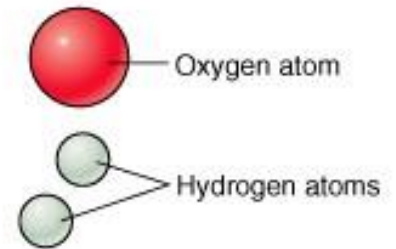
Minerals & Rocks



Matter – Water

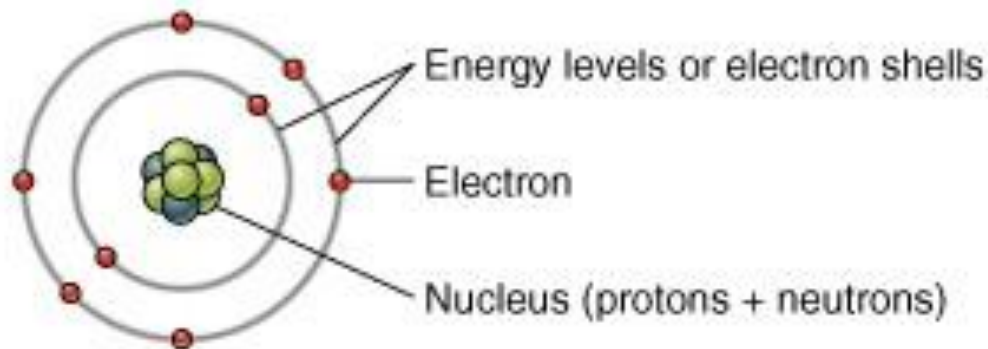


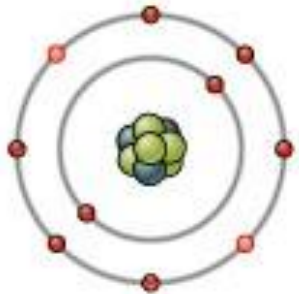
Molecules of water
containing the chemical
elements hydrogen and oxygen



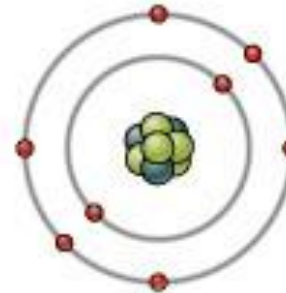
Elements made up of atoms

ATOM





Two electrons (●) have been added to completely fill the outer energy level. This means that the atom is now an ion with a -2 charge because it has 2 more electrons than protons.



Atomic mass number
Chemical symbol for the element
Atomic number

ANION, KATION, MASA DAN NOMER ATOM

ISOTOPE: ATOM DGN JUMLAH NEUTRON BERBEDA

Carbon 12 (^{12}C)
6 protons + 6 neutrons



98.9% of
all carbon atoms

Carbon 13 (^{13}C)
6 protons + 7 neutrons



1.1% of
all carbon atoms

Carbon 14 (^{14}C)
6 protons + 8 neutrons



0.0000000001% of
all carbon atoms

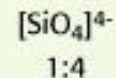
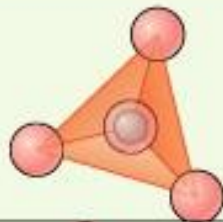
● Protons
● Neutrons

TABLE EM2.2-1

Group	Schematic Arrangement of Silica Tetrahedra	Unit Composition and silicon: oxygen ratio	Example mineral and formula
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Nesosilicates

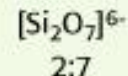
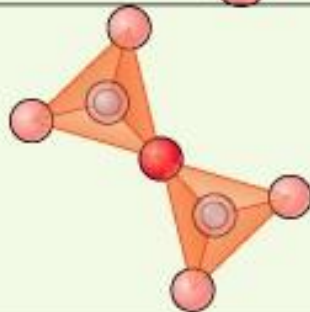
Isolated tetrahedra bonded to positive ions. There is no sharing of oxygen ions between tetrahedra.



Olivine
 $(\text{Mg,Fe})_2\text{SiO}_4$

Sorosilicates

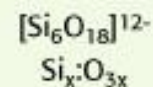
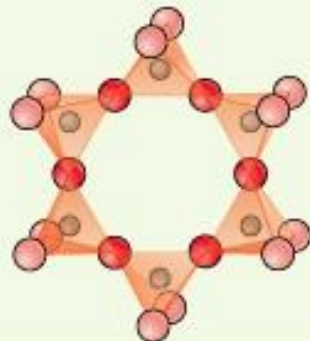
Two silica tetrahedra linked with one shared oxygen ion. Other positive ions bond to remaining oxygen ions at unshared tetrahedra corners.



Epidote
 $\text{Ca}_2(\text{Fe,Al})\text{Al}_2\text{O}(\text{SiO}_4)(\text{Si}_2\text{O}_7)(\text{OH})$

Cyclosilicates

Three or more silica tetrahedra linked into closed rings. Each tetrahedron shares oxygen ions with two adjacent tetrahedra. Other positive ions bond to unshared oxygen ions.



Beryl
 $\text{Be}_3\text{Al}_2\text{Si}_6\text{O}_{18}$

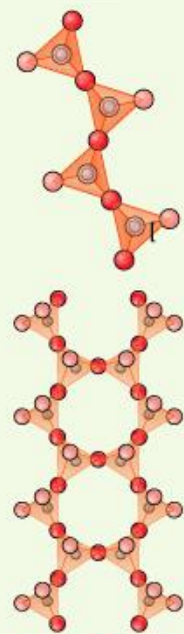
SILIKAT STRUKTUR-2

Inosilicates

Infinitely linked tetrahedra forming single or double chains. In single chains two oxygen ions from each silica tetrahedra are shared.

Double chains are two single chains linked to each other where either 2-3 oxygen ions of each silica tetrahedron are shared.

Other positive ions bond to unshared oxygen ions to link the chains together. The hydroxyl ion (OH^-), fits loosely in the openings between the two tetrahedral chains.



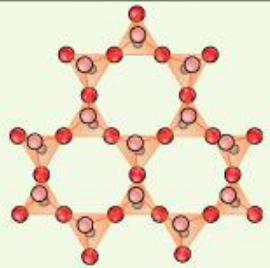
Pyroxene
 MgSiO_3



Amphibole
 $(\text{Ca}, \text{Na})_{2-3}(\text{Mg}, \text{Fe}, \text{Al})_5$
 $\text{Si}_8\text{O}_{22}(\text{OH})_2$

Phyllosilicates

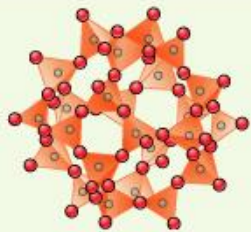
Also known as sheet silicates. The three oxygen ions at the base of the silica tetrahedra are all shared, producing infinitely flat sheet in two-dimensions. Other positive ions bond to unshared oxygen ions.



Muscovite
 $\text{KAl}_2(\text{AlSi}_3\text{O}_{10})(\text{OH})_2$

Tectosilicates

Also known as the framework silicates. All four oxygen ions of the silica tetrahedra are shared to form a three-dimensional framework. Other positive ions bond to the structure to balance charges that result from Al^{3+} substitution for Si^{4+} .



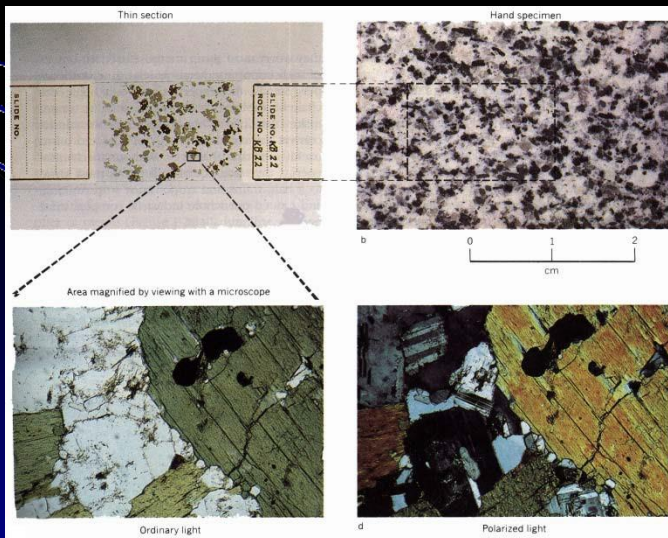
Quartz
 SiO_2

Earth Materials

Minerals & Rocks



Crystals & Minerals



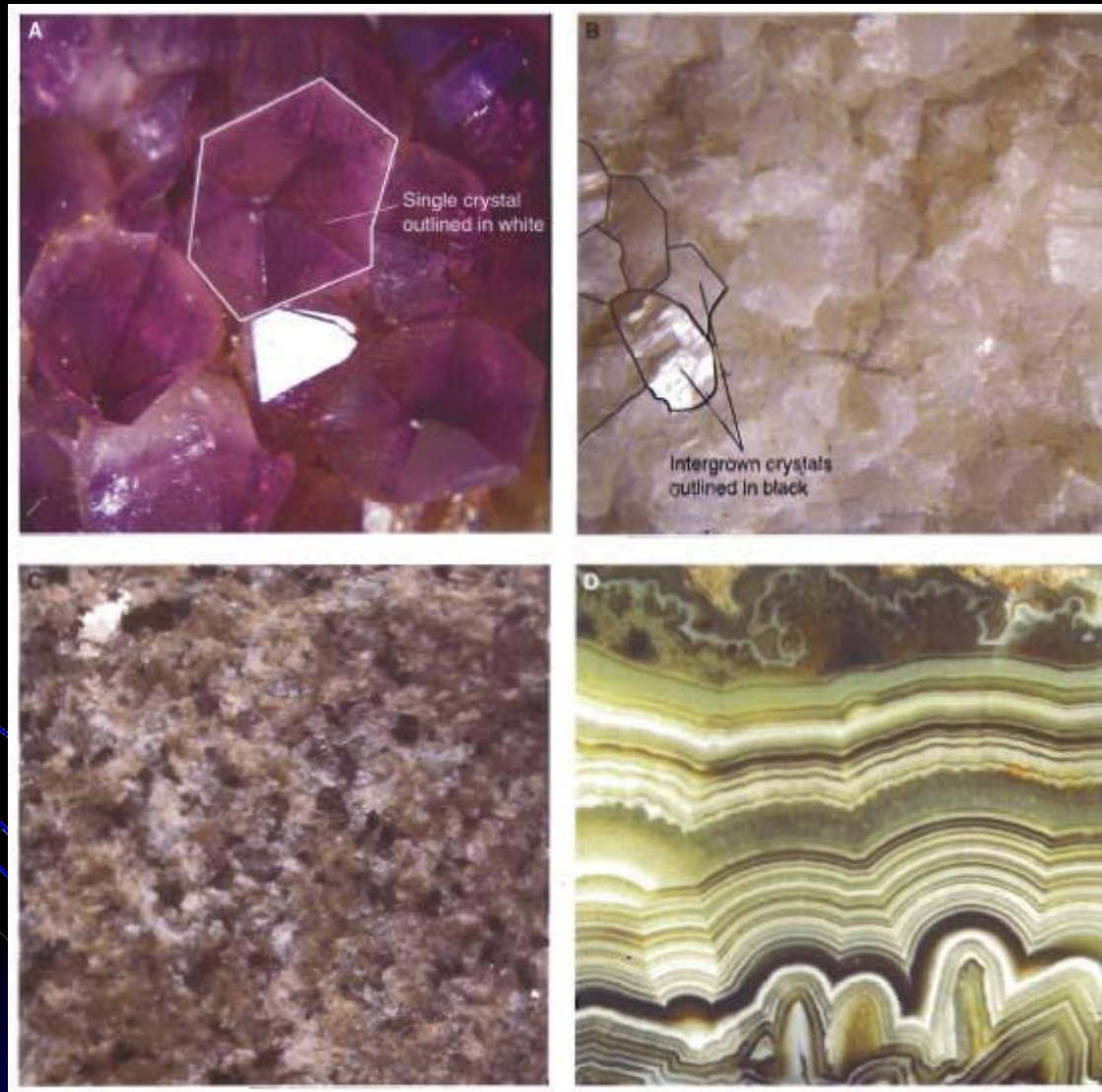
Microscopic view

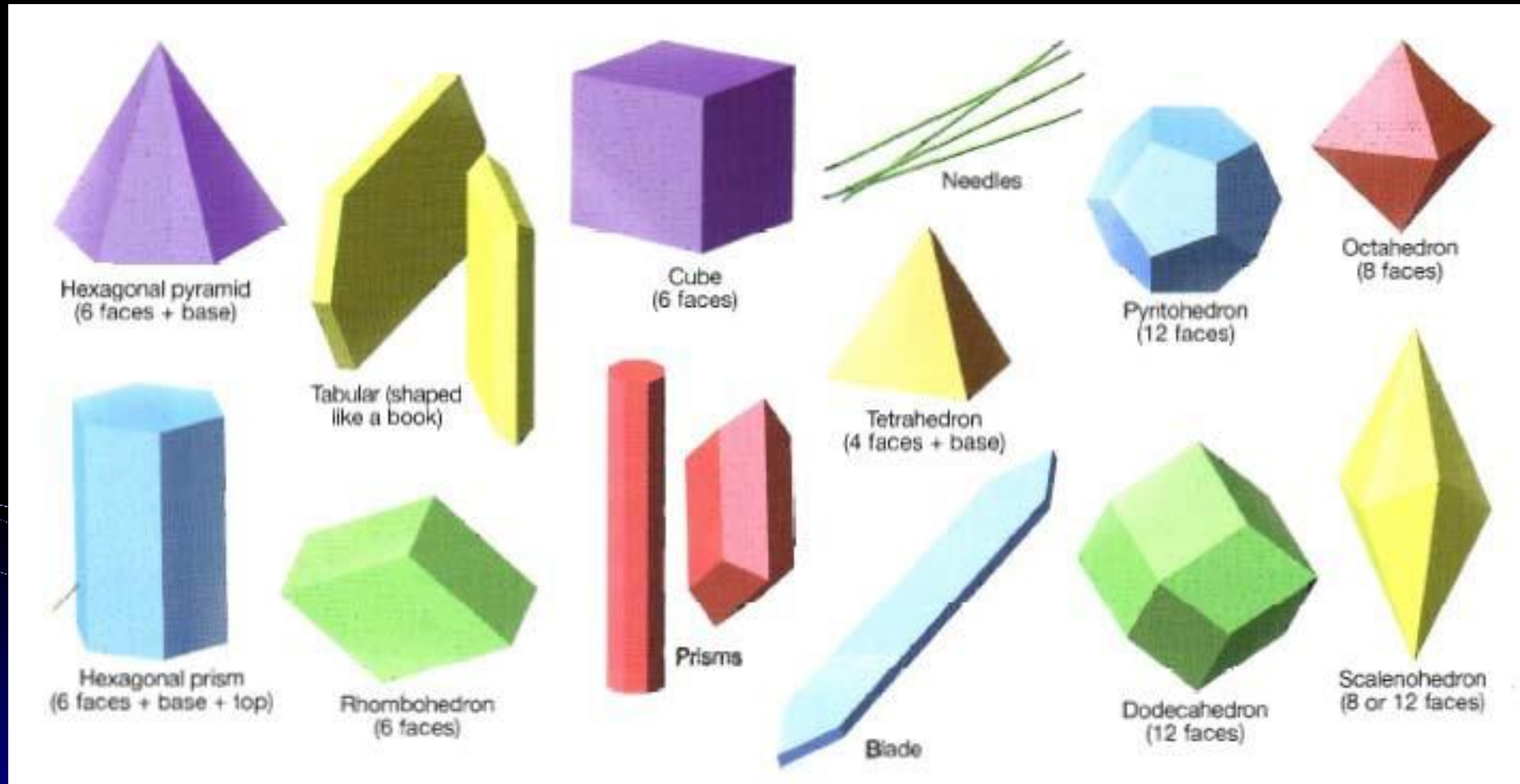


Mineral dan Batuan

Mineral dan Batuan

- **Komposisi dan struktur mineral membentuk sifat fisik seperti warna, kekerasan, *cleavage*, bentuk kristal dan *specific gravity*.**
- **Beberapa mineral termasuk sangat umum dijumpai pada batuan sehingga dikenal sebagai mineral pembentuk batuan (*rock-forming*). Kebanyakan dari jenis ini adalah silikat, tetapi beberapa dari group karbonat juga sangat penting terutama dalam batuan sedimen.**
- **Sumberdaya mineral adalah tempat terkonsentrasinya mineral ekonomis yang penting. Cadangan adalah konsentrasi dari sumberdaya mineral yang mempunyai nilai ekonomis.**
- **Catatan: Persentasi volume dari 8 elemen utama: Oksigen (93.8%), Potassium (1.8%), Sodium (1.3%), Calcium (1.0%), Silikon (0.9%), Aluminum (0.5%), Iron (0.4%), Magnesium (0.3%)**

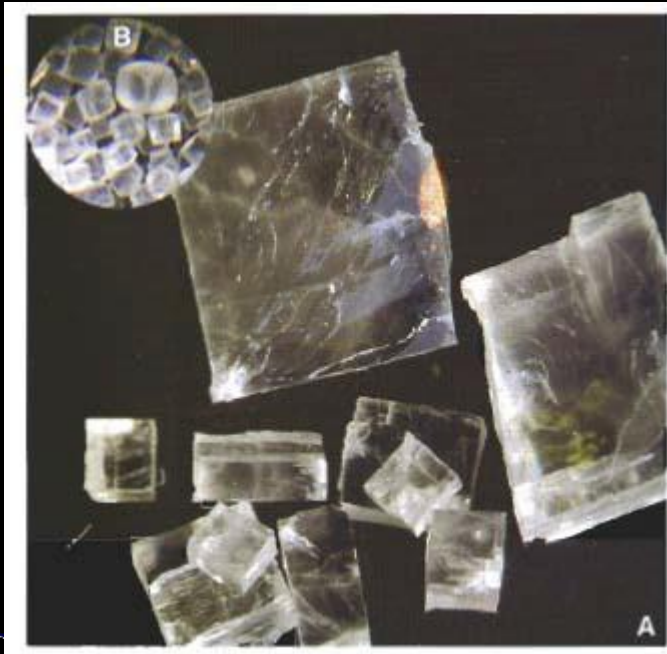




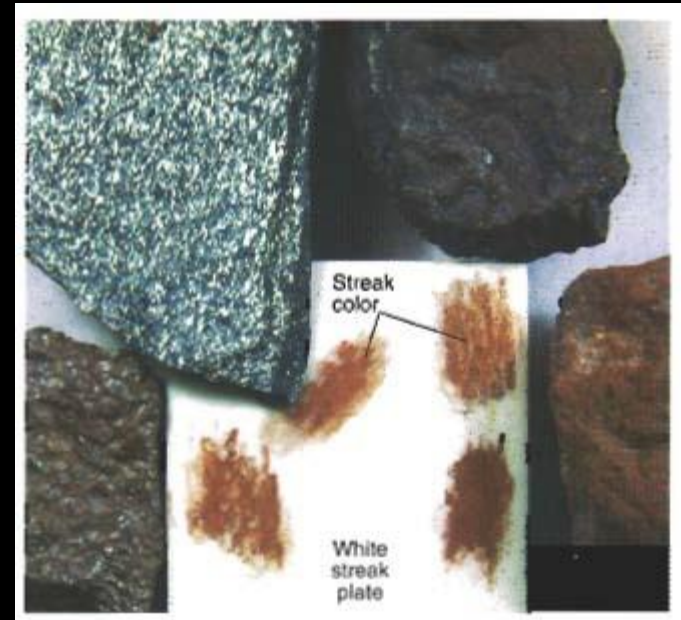
Crystal Forms

Mineral dan Batuan

Kristal & Mineral



HALITE


















HEMATITE

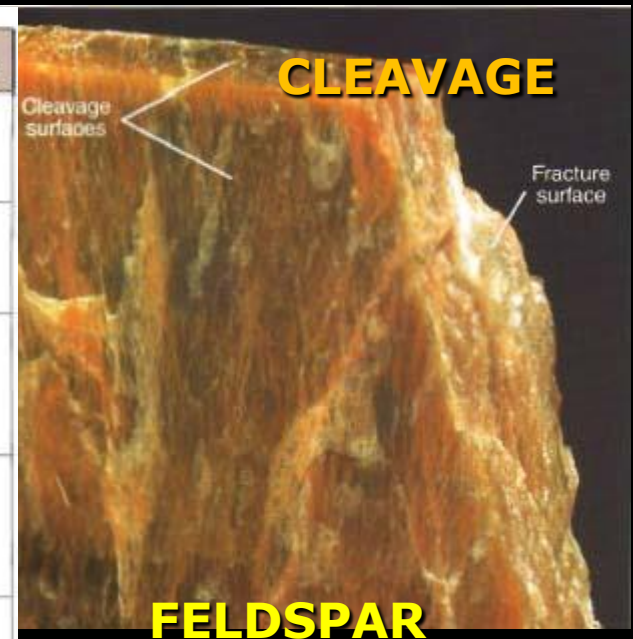
- Crystal Form
- Color
- Lustre
- Streak
- Cleavage
- Hardness

Physical Properties

Mineral dan Batuan

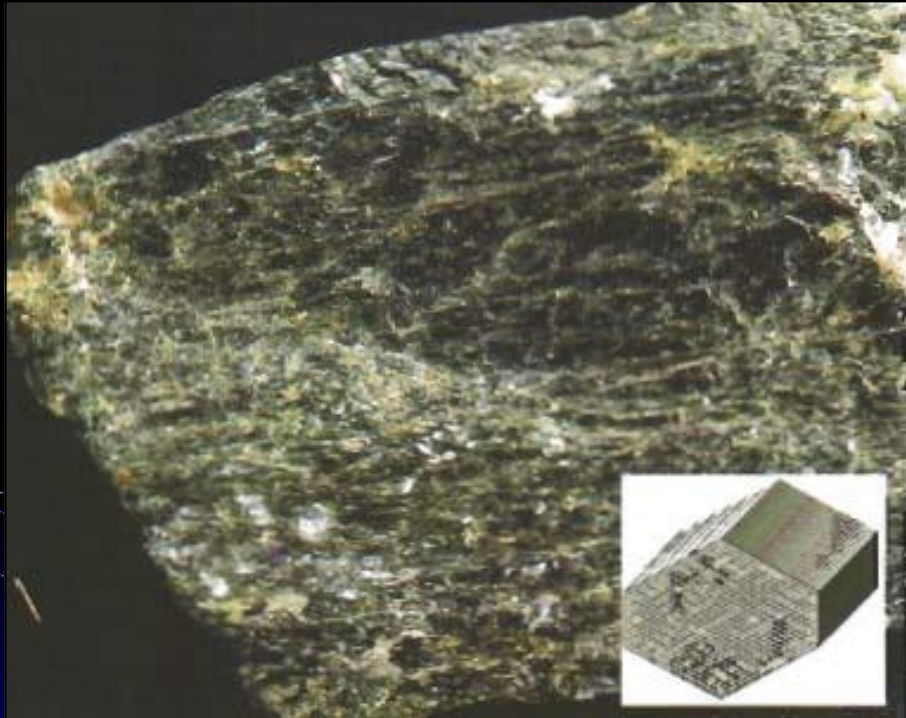
Kristal & Mineral

Number of Cleavage Directions	Shapes that Crystal Breaks into	Sketch	Illustration of Cleavage Directions
0 No cleavage, only fracture	Irregular masses with no flat surfaces		None
1	Basal cleavage "Books" that split apart along flat sheets		
2 at 90°	Elongated form with rectangular cross sections (prisms) and parts of such forms		
2 not at 90°	Elongated form with parallelogram cross sections (prisms) and parts of such forms		
3 at 90°	Cubic cleavage Shapes made of cubes and parts of cubes		
3 not at 90°	Rhombohedral cleavage Shapes made of rhombohedrons and parts of rhombohedrons		
4	Octahedral cleavage Shapes made of octahedrons and parts of octahedrons		
6	Shapes made of dodecahedrons and parts of dodecahedrons		

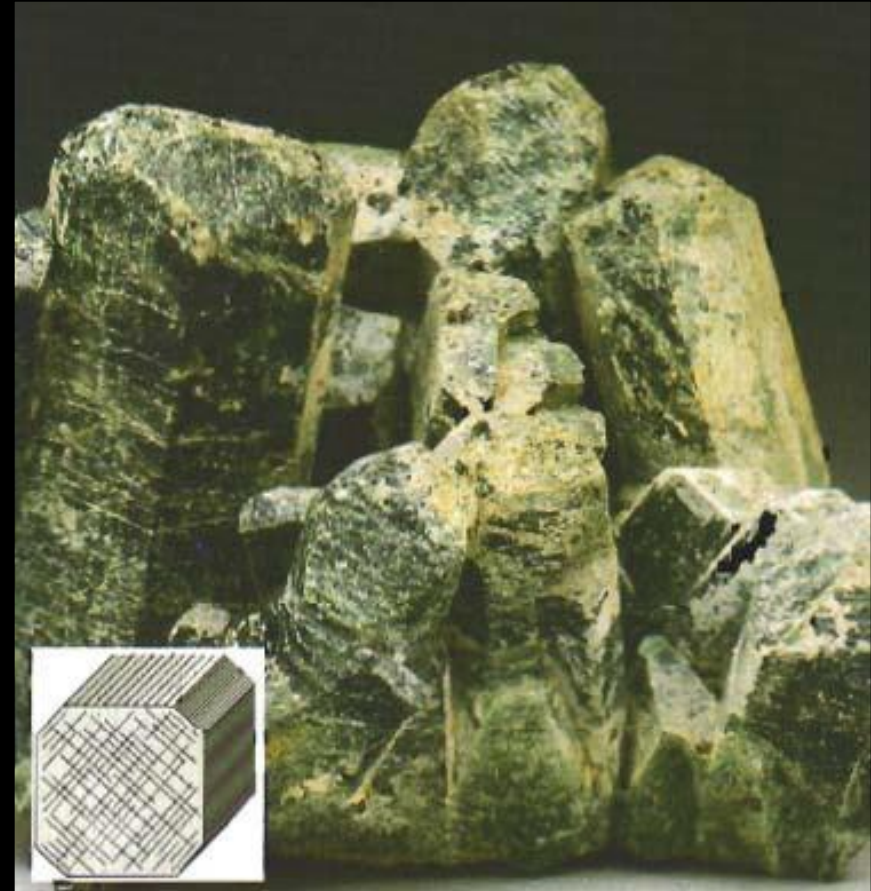


Mineral dan Batuan

Kristal & Mineral



HORNBLEND


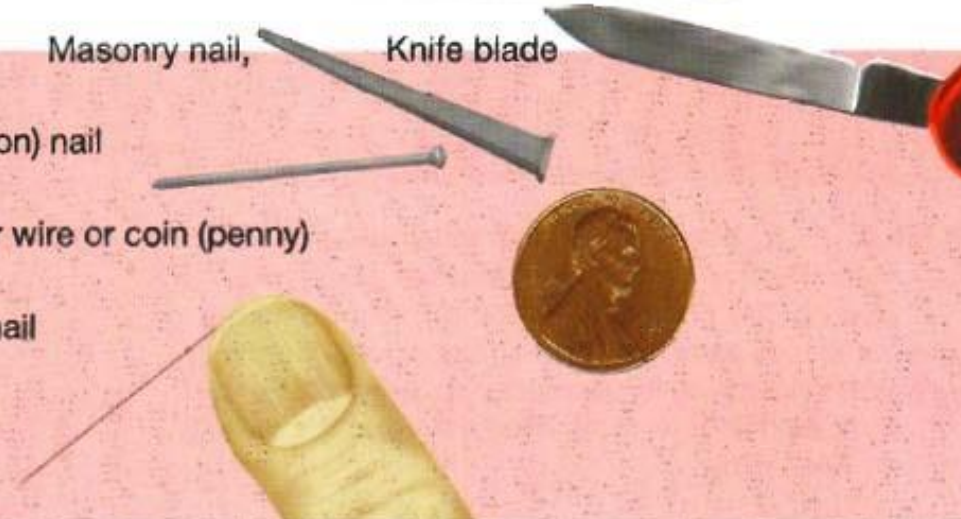


PYROXENE

CLEAVAGE

Mineral dan Batuan

Kristal & Mineral

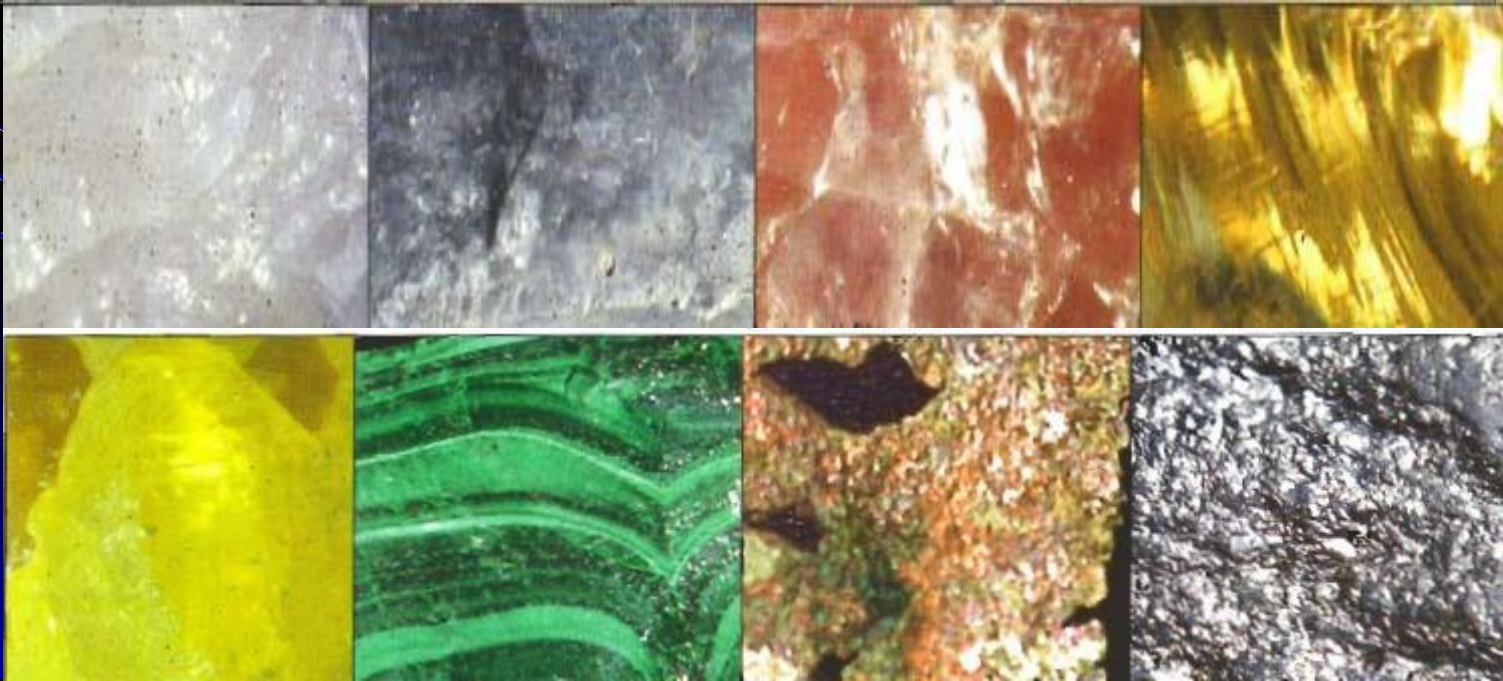
Mohs Scale of Hardness*		Hardness of Some Common Objects	
HARD	10 Diamond		
	9 Corundum		
	8 Topaz		
	7 Quartz		
	6 Orthoclase Feldspar <i>Orthoclase</i>		
SOFT	5 Apatite	5.5 Glass, Masonry nail, Knife blade	
	4 Fluorite	4.5 Wire (iron) nail	
	3 Calcite	3.5 Copper wire or coin (penny)	
	2 Gypsum	2.5 Fingernail	
	1 Talc		

SKALA MOHS



- **CALCITE DAN HCL**
- **WARNA**

SIFAT KHUSUS



Mineral dan Batuan

Kristal & Mineral

MINERAL DATABASE (Alphabetical Listing)					
Mineral	Luster	Hardness	Streak	Distinctive Properties	Some Uses
ACTINOLITE (amphibole)	Nonmetallic (NM)	5.5-6	White	Color dark green or pale green; Forms needles, prisms, and asbestos fibers; Good cleavage at 60° and 120°; SG = 3.1	Gemstone (Nephrite), Asbestos products
AMPHIBOLE: See HORNEBLINDE and ACTINOLITE					
APATITE $\text{Ca}_5(\text{FPO}_3)_3$, calcium fluorophosphate	Nonmetallic (NM)	5	White	Color pale or dark green, brown, blue, white, or purple; Sometimes colorless; Transparent or opaque; Brittle; Conchoidal fracture; Forms hexagonal prisms; SG = 3.1-3.4	Used for pesticides and fertilizers
ASBESTOSE: fibrous varieties of AMPHIBOLE and SERPENTINE					
AUGITE (pyroxene) calcium ferromagnesian silicate	Nonmetallic (NM)	5.5-6	White to pale gray	Color green to black; Forms opaque, short, 8-sided prisms; Two good cleavages that intersect at 60° and 90° (nearly right angles); SG = 3.2-3.5	Some pyroxene mined as an ore of titanium, for making steel
AZURITE $\text{Cu}_3(\text{OH})_2(\text{CO}_3)_2$, hydrous copper carbonate	Nonmetallic (NM)	3.5-4	Light blue	Color a distinctive deep blue; Forms crusts of small crystals, opaque earthy masses, or short and long prisms; Brittle; Effervesces in dilute HCl; SG = 3.7-3.8	One of copper for pipes, electrical circuits, coins, ammunition, gemstone
BARITE BaSO_4 , barium sulfate	Nonmetallic (NM)	3-3.5	White	Colorless to white, with tints of brown, yellow, blue, or red; Forms short tabular crystals and rose-shaped masses (Barite roses); Brittle; Cleavage good to excellent; Very heavy; SG = 4.5	Used in rubber, paint, glass, oil-well drilling fluids
BAUXITE (mixture of aluminum hydroxides)	Nonmetallic (NM)	1-3	White	Brown earthy rock with streaks of gray, white, and yellow; Amorphous; Often contains rounded pea-sized structures with laminations; SG = 2.3	One of Aluminum
BIOTITE MICA ferromagnesian potassium, hydrous aluminum silicate	Nonmetallic (NM)	2.5-3	Gray-brown	Color black, green-black, or brown-black; Cleavage excellent; Forms very short prisms that split easily into very thin, flexible sheets; SG = 2.7-3.1	Used for fire-resistant tiles, rubber, paint
BORNITE Cu_5FeS_4 , copper-iron sulfide	Metallic (M)	3	Dark gray to black	Color opaque silvery blue or copper-red; Tarnishes to iridescent purple and blue; Forms dense, friable masses; Cleavage poor to absent	One of copper for pipes, electrical circuits, coins, ammunition, brass, bronze
CALCITE CaCO_3 , calcium carbonate	Nonmetallic (NM)	3	White	Usually colorless, white, or yellow, but may be green, brown, or pink; Opaque or transparent; Excellent cleavage in 3 directions not at 90°; Forms prisms, rhombohedrons, or scalenohedrons that break into rhombohedrons; Effervesces in dilute HCl; SG = 2.7	Used to make antacid tablets, fertilizer, cement; One of calcium
CHALCEDONY SiO_2 , cryptocrystalline quartz	Nonmetallic (NM)	7	White	Colorless, white, yellow, light brown, or other pastel colors in banded bands; Often translucent; Conchoidal fracture; Luster waxy; Cryptocrystalline; SG = 2.3-2.6	Used as an abrasive; Used to make glass, gemstones (agate, chrysoprase)

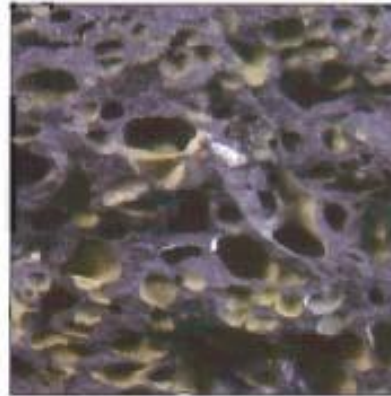
MINERAL DATABASE (Alphabetical Listing)					
Mineral	Luster	Hardness	Streak	Distinctive Properties	Some Uses
CHALCOPYRITE CuFeS_2 , copper-iron sulfide	Metallic (M)	3.5-4	Dark gray	Color golden or brassy yellow; Tarnishes brown, or iridescent blue, green, and red; Forms elongate tetrahedra; Brittle; Cleavage poor; SG = 4.1-4.3	One of copper for pipes, electrical circuits, coins, ammunition, brass, bronze
CHERT SiO_2 , cryptocrystalline quartz	Nonmetallic (NM)	7	White	Opaque gray or white; Luster waxy; Conchoidal fracture; SG = 2.5-2.8	Used as an abrasive; Used to make glass, gemstones
CHLORITE ferromagnesian aluminum silicate	Nonmetallic (NM)	2-2.5	White	Color dark green; Cleavage excellent; Forms short prisms that split easily into thin flexible sheets; Luster shiny or dull; SG = 2-3	Used for fire-resistant tiles, rubber, paint, art sculpture medium
CHROMITE FeCr_2O_4 , iron-chromium oxide	Metallic (M)	5.5-6	Dark brown	Color shaly black to black; Tarnishes gray; Forms octahedra; Brittle; No cleavage; May be weakly magnetic; SG = 4.6-4.8	One of chromium for making chrome, stainless steel, mirrors, paint and used in leather tanning
CHRYSOCOLLA $\text{CuSiO}_3 \cdot 2\text{H}_2\text{O}$, hydrated copper silicate	Nonmetallic (NM)	2-4	Very light blue	Color pale blue to blue-green; Opaque; Forms amorphous crusts or may be massive; Conchoidal fracture; Luster shiny or earthy; SG = 2.0-2.4	One of copper for pipes, electrical circuits, coins, ammunition, gemstone
COPPER (NATIVE COPPER) Cu , copper	Metallic (M)	2.5-3	Copper	Color copper; Tarnishes brown or green; Forms distorted cubes octahedrons, and dendritic foot-like masses; Malleable; Opaque; Cleavage absent; SG = 8.8-8.9	One of copper for pipes, electrical circuits, coins, ammunition, brass, bronze
CORUNDUM Al_2O_3 , aluminum oxide	Nonmetallic (NM)	9	White	Color gray, blue, red, brown; Transparent or opaque; Forms short hexagonal prisms with striated flat ends; Cleavage absent; SG = 3.9-4.1	Used for abrasive powders to polish lenses; gemstones (red ruby, blue sapphire)
DOLomite $\text{CaMg}(\text{CO}_3)_2$, magnesium calcium carbonate	Nonmetallic (NM)	3.5-4	White	Color white, gray, brown, or pink; Usually opaque; Cleavage excellent in 3 directions; Breaks into rhombohedrons; Resolves in dilute HCl only if powdered; SG = 2.8-2.9	One of magnesium metal; soft abrasive; used to make paper
EPIDOTE complex silicate	Nonmetallic (NM)	6-7	White	Color pale or dark green to yellow-green; Massive or forms striated prisms; Cleavage poor; SG = 3.3-3.5	Gemstone
FELDSPAR: See PLAGIOCLASE (Na-Ca Feldspars) and POTASSIUM FELDSPAR (K-Feldspar)					
FLINT SiO_2 , cryptocrystalline quartz	Nonmetallic (NM)	7	White	Color black to very dark gray; Opaque to translucent; Conchoidal fracture; Cryptocrystalline; SG = 2.5-2.6	Used as an abrasive; Used to make glass, gemstones
FLUORITE CaF_2 , calcium fluoride	Nonmetallic (NM)	4	White	Colorless, purple, blue, gray, green, or yellow; Crystals usually cubic; Transparent or opaque; Brittle; SG = 3.0-3.3	Source of fluorine for processing aluminum; flux in steel making

DATABASE MINERAL

COMMON IGNEOUS ROCK TEXTURES



A Glassy



B Vesicular (bubbly)



C Randomly oriented small crystals



D Randomly oriented large crystals

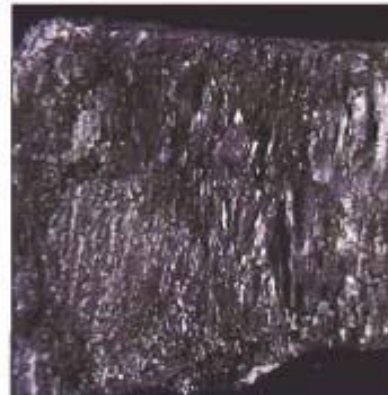
COMMON METAMORPHIC ROCK TEXTURES



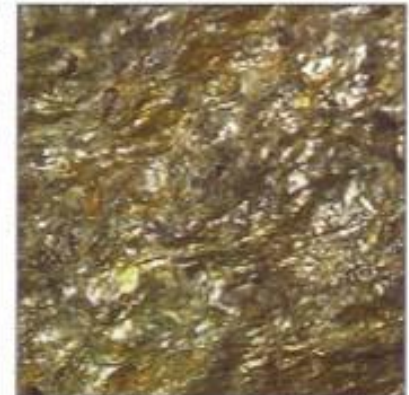
E Deformed fossils



F Folded



G Oriented small crystals and shiny reflection



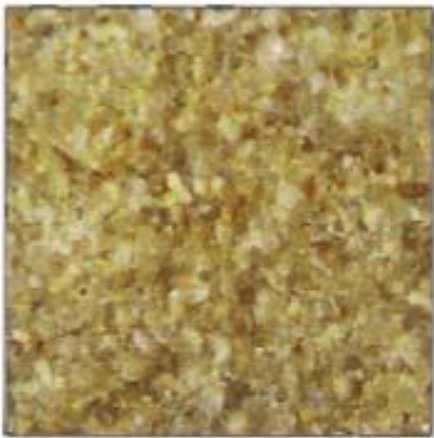
H Oriented equigranular large crystals



COMMON SEDIMENTARY ROCK TEXTURES



I Silty clastic
(note fossil)



J Sandy clastic



K Layered
small crystals



L Coarse-grained
clastic



Figure 1.5 Quartz crystal.



Figure 1.6 Rose quartz.



Figure 1.7 Smoky quartz.



Figure 1.8 Cryptocrystalline quartz (chert).



Figure 1.9 Orthoclase (microcline).



Figure 1.10 Plagioclase.



Figure 1.11 Gypsum (selenite).



Figure 1.12 Talc.



Figure 1.13 Calcite (note double refraction).



Figure 1.14 Fluorite.



Figure 1.15 Biotite.



Figure 1.16 Olivine.



Figure 1.17 Hematite (specularite).



Figure 1.18 Hematite.



Figure 1.19 Goethite (limonite).



Figure 1.20 Pyrite.

THE ORIGIN OF ROCKS



Igneous Rocks

Origin

PLATE TECTONICS & ORIGIN OF ROCKS

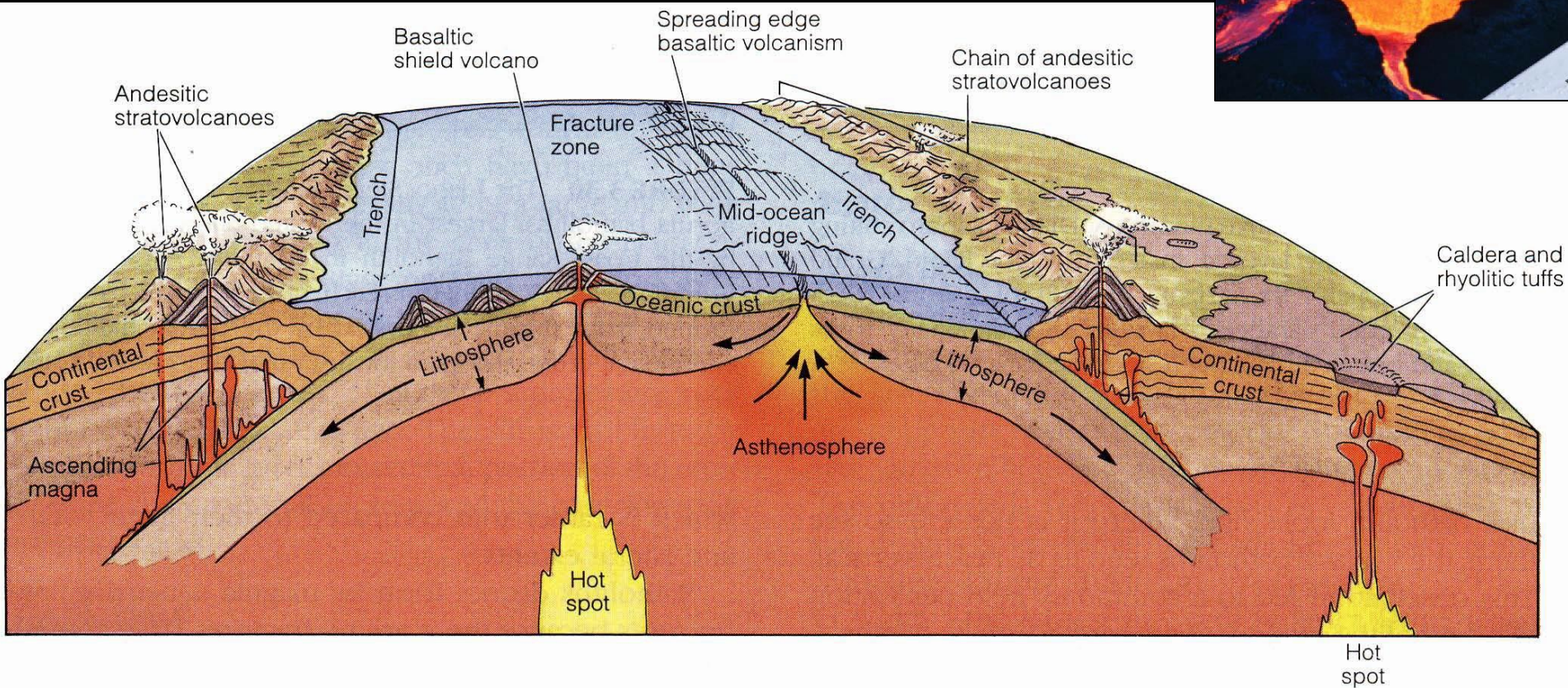
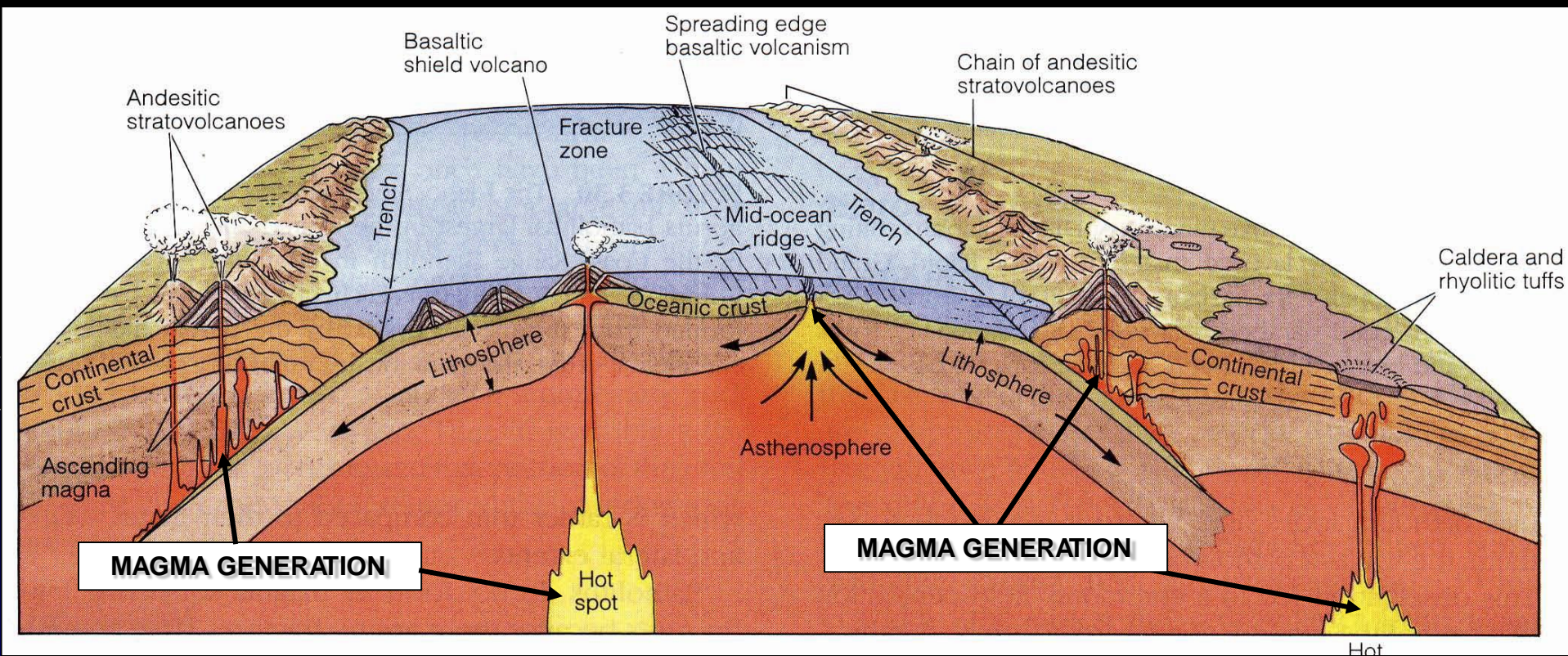


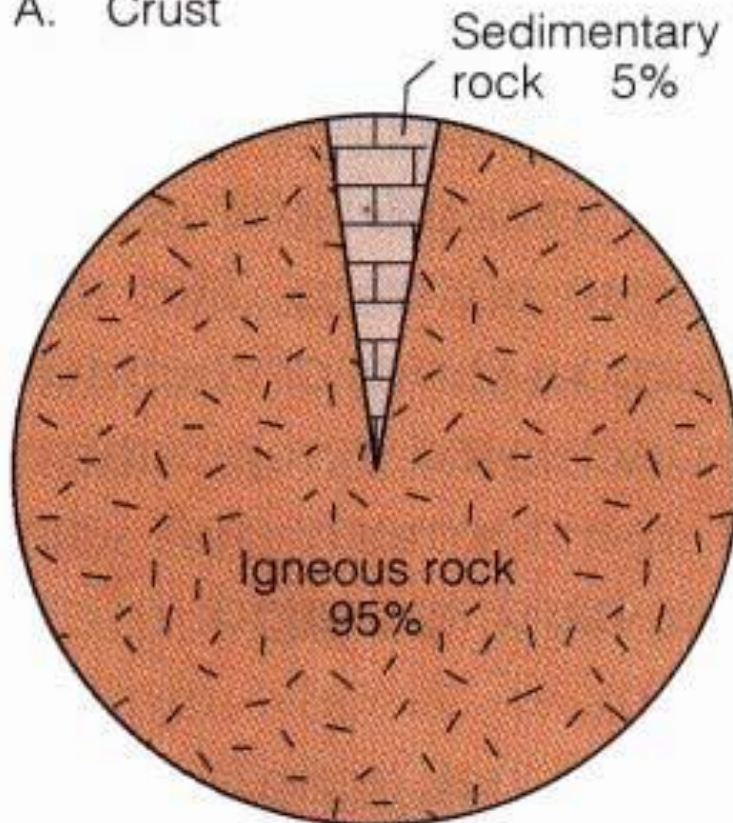
PLATE TECTONICS & ORIGIN OF ROCKS



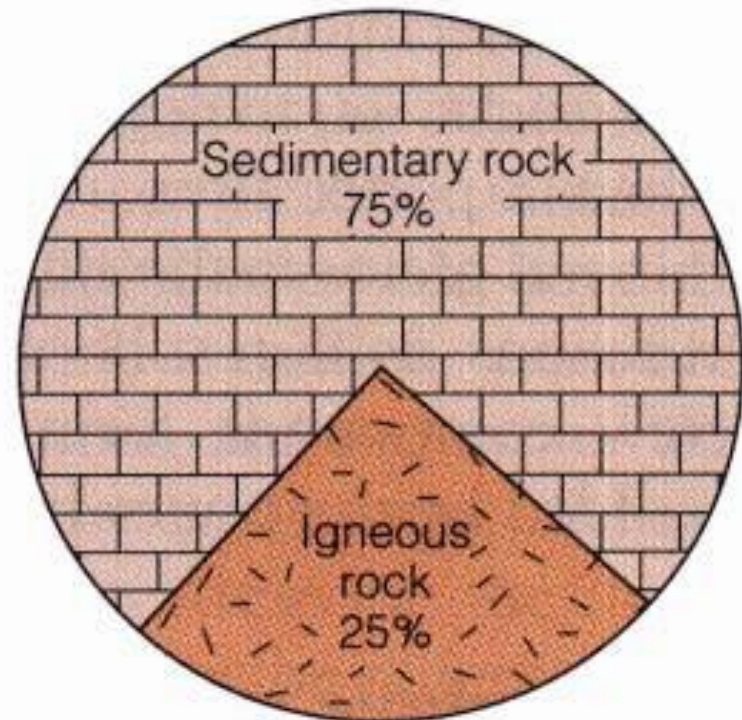
Earth Crust

Sedimentary vs Ign Rocks

A. Crust



B. Surface

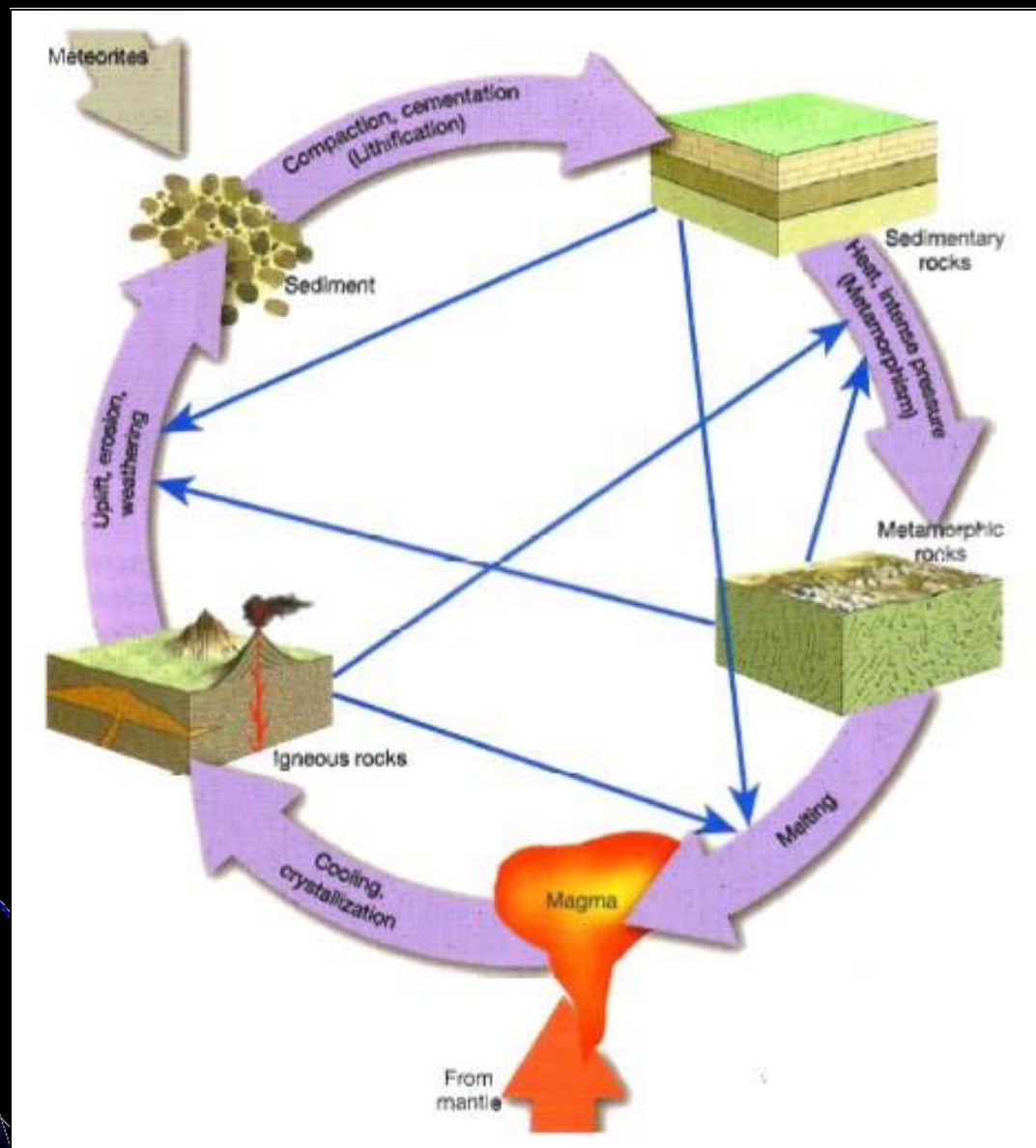


Rock distribution

Earth Crust

Rocks Cycle

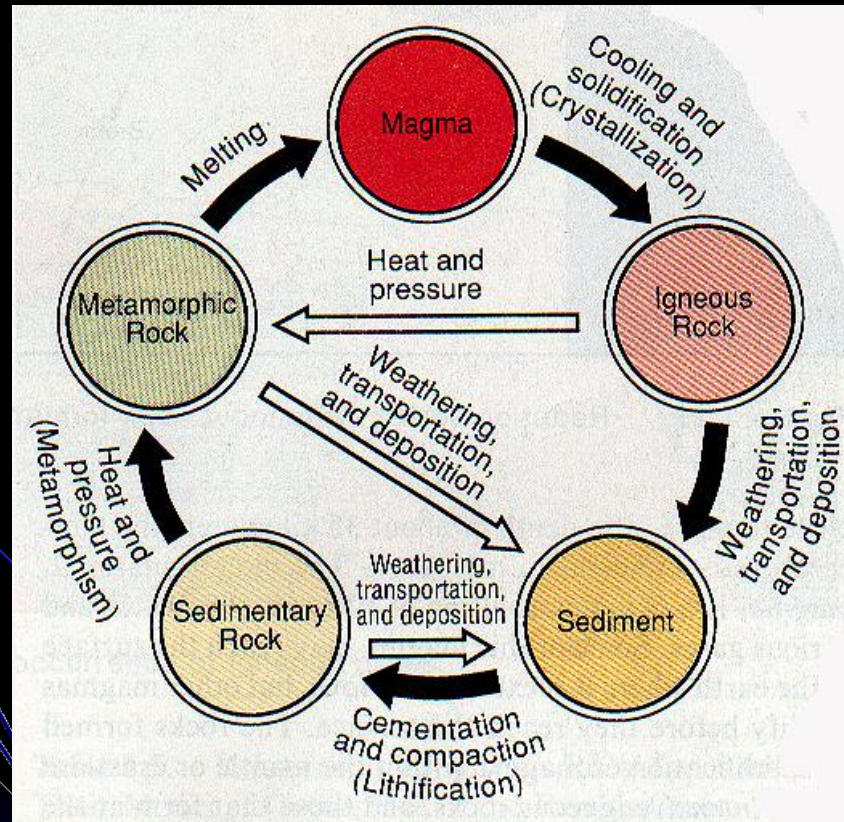
ROCK CYCLES



Earth Crust

Rocks Cycle

Siklus Batuan menggambarkan proses interaksi dan transformasi dari tiga kelompok batuan. Proses ini dikontrol oleh proses internal seperti tektonik dan eksternal yaitu udara dan air.



Earth Crust

Rocks Cycle

Sedimentary Rock

Extrusive Igneous Rock

Sedimentary Rock

volcano

Sedimentary Basin

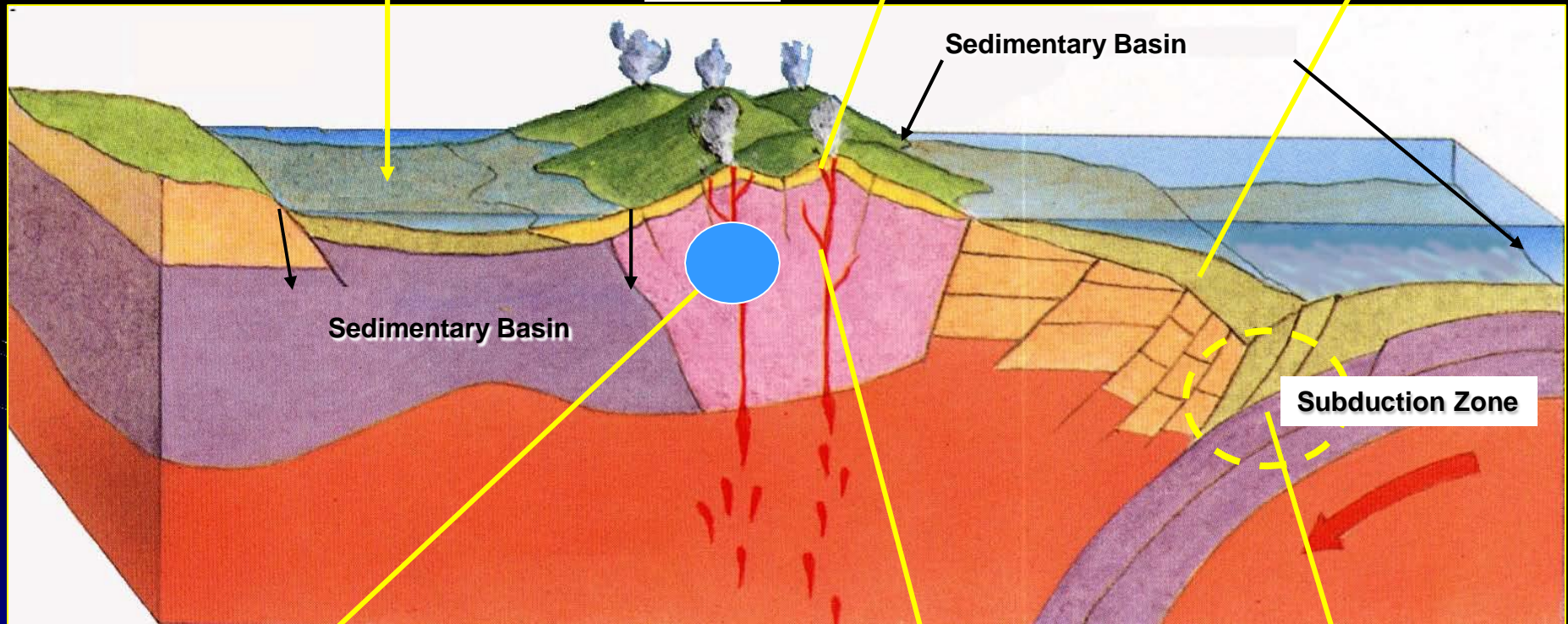
Sedimentary Basin

Subduction Zone

**Thermal
Metamorphic Rock**

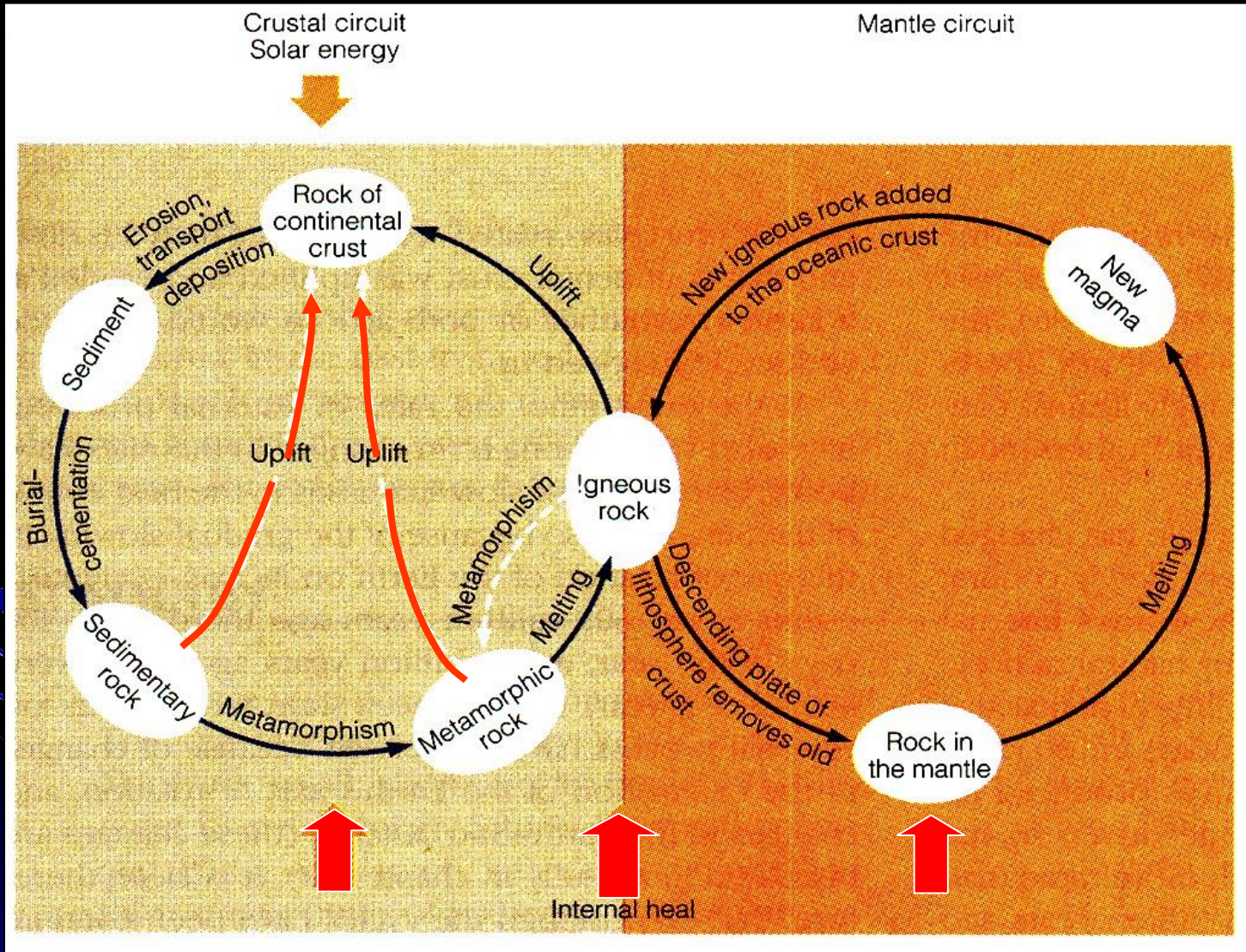
Intrusive Rocks

**Dinamo-Thermal
Metamorphic Rock**



Earth Crust

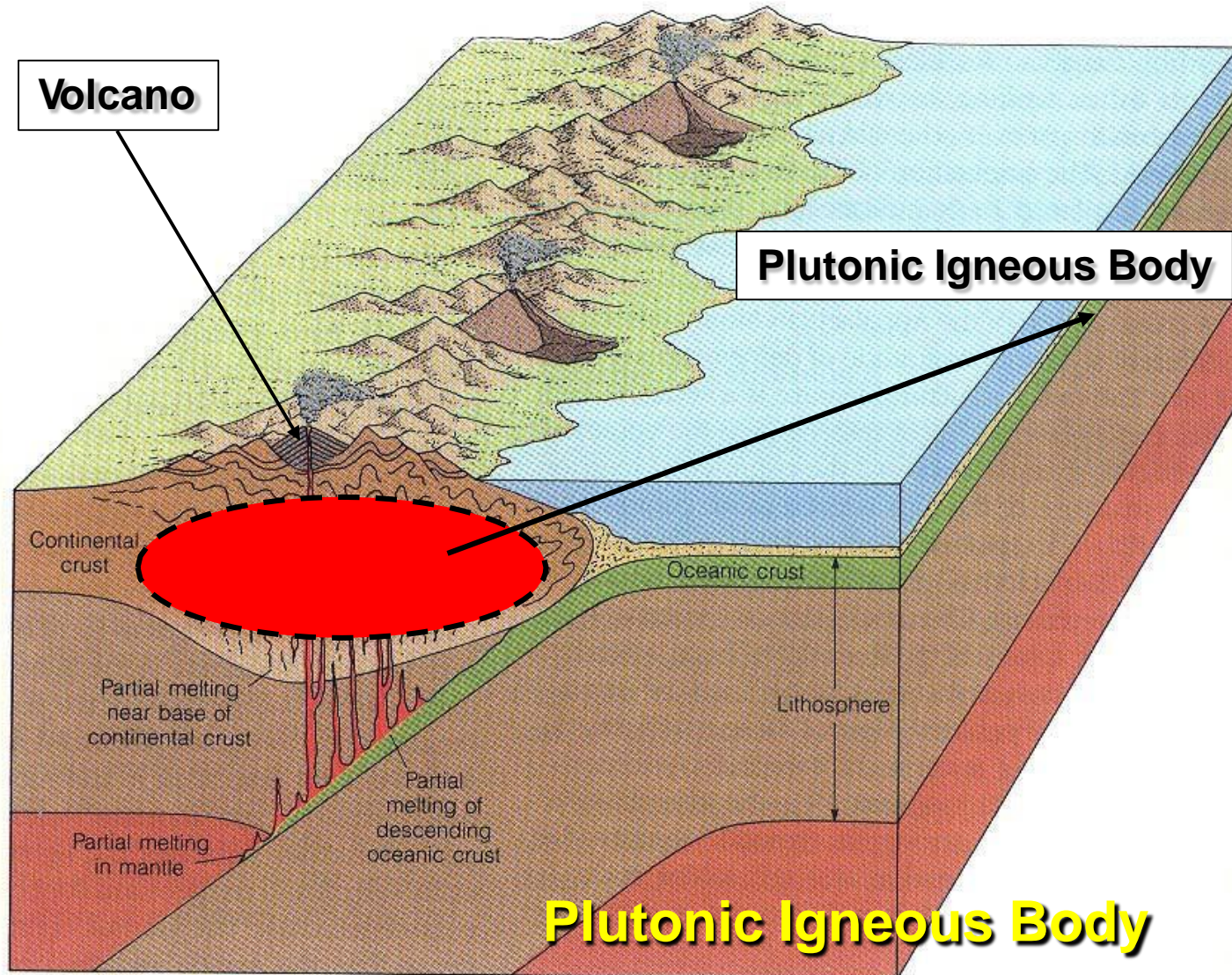
Rocks vs Tectonic Cycle



Rock Cycle & Plate Tectonics

Igneous Rocks

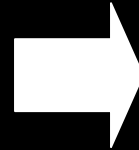
Plutonic vs Volcanic



Igneous Rocks

Plutonic vs Volcanic

EXTRUSI VE (VOLCANIC)



PIROCLASTIC

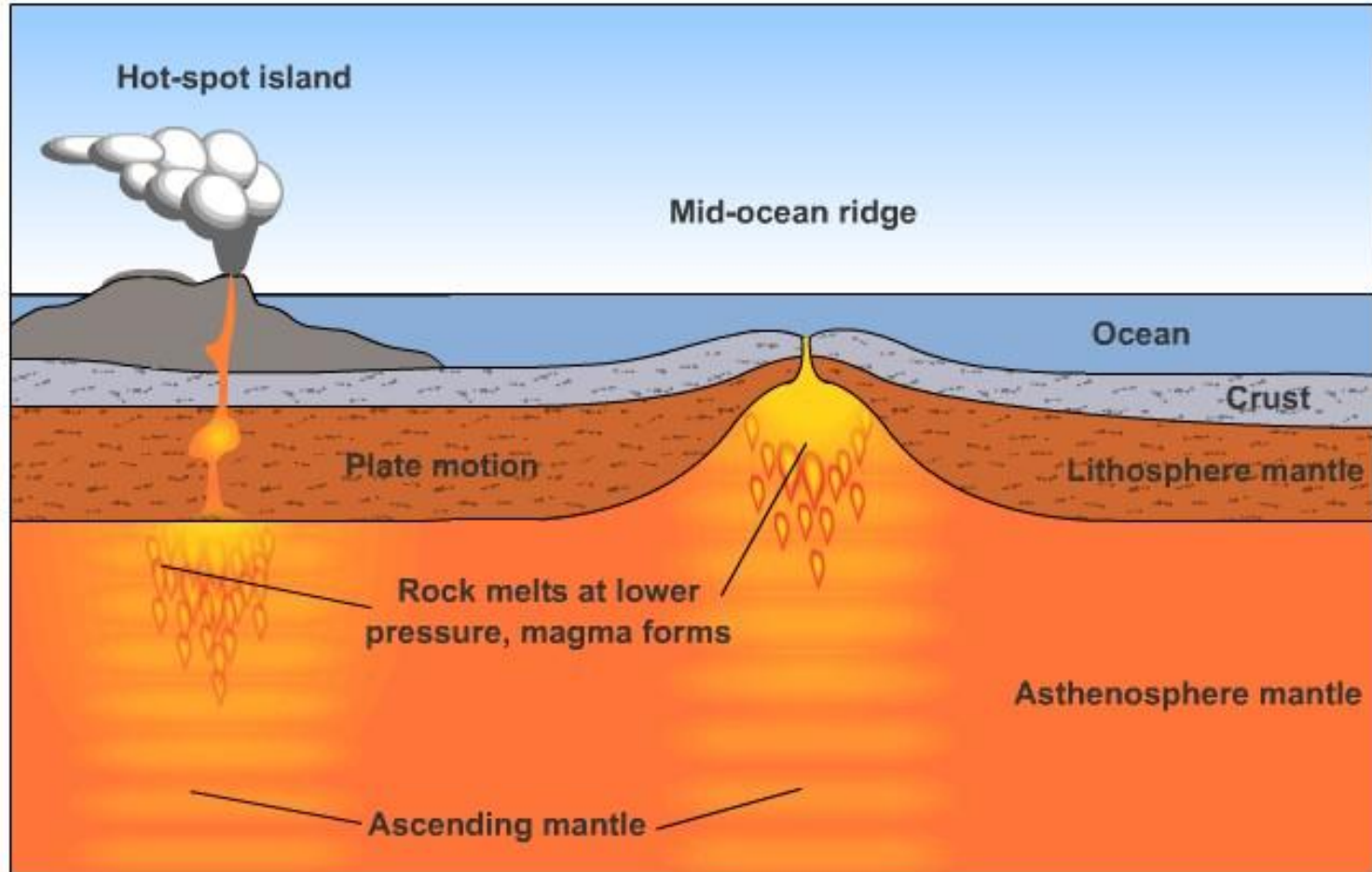


Lava Flow



Origin of Igneous Rocks

Plate Tectonics and mantle melting - Generating magma by lowering pressure

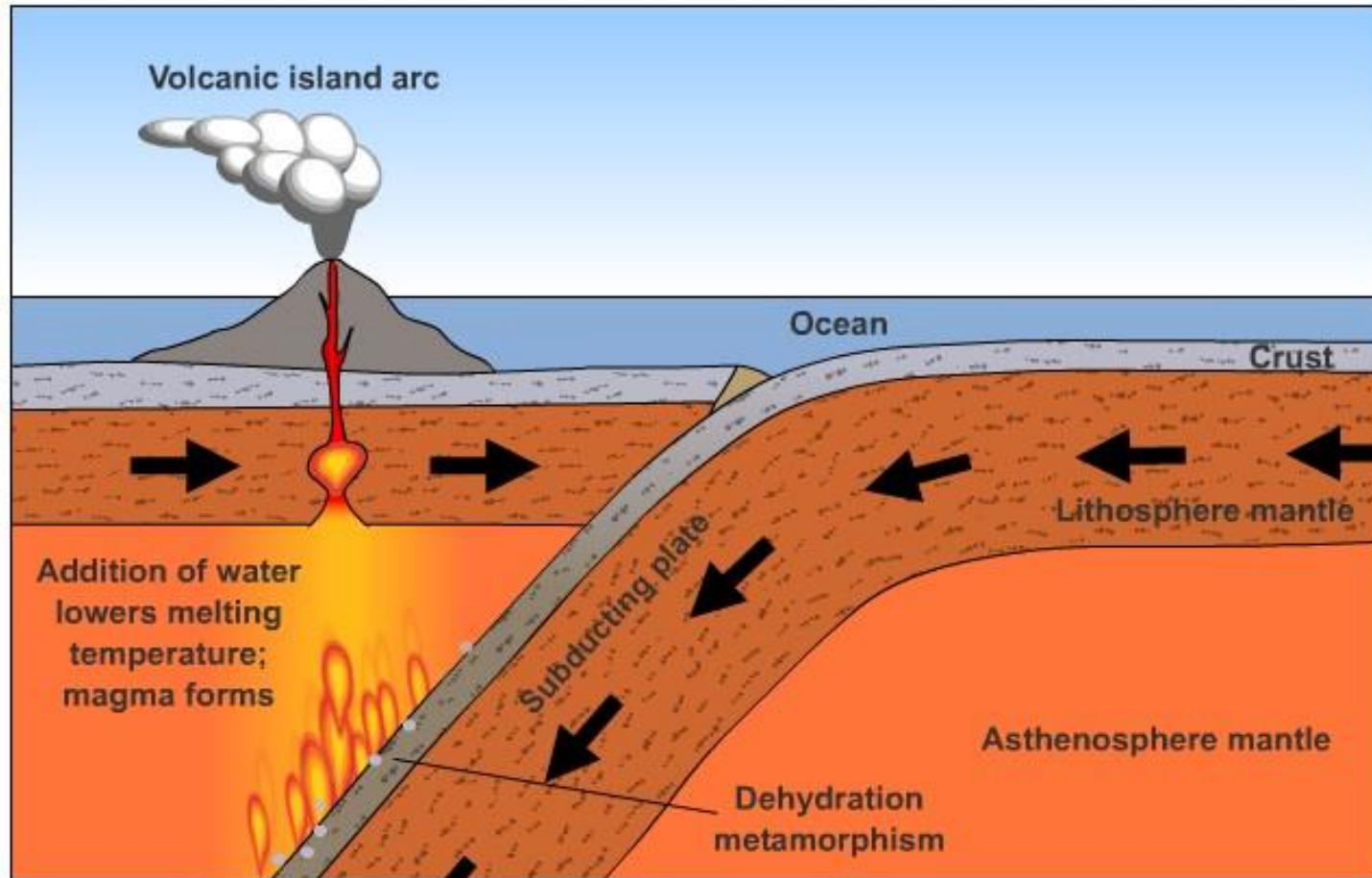


MAGMA GENERATION IN SPREADING CENTER

Igneous Rocks

Plutonic vs Volcanic

Plate Tectonics and mantle melting - Generating magma by adding water



MAGMA GENERATION IN SUBDUCTION ZONE



Igneous Rocks

Plutonic vs Volcanic

- Mineral diawali kristalisasi dari magma dan lava sesudah inti kristal kecil terbentuk dan berkembang.
- Pendinginan cepat yang umumnya terjadi pada batuan vulkanik dan membentuk batuan yang bertekstur **afanitik (berbutir halus)**.
- Pendinginan perlahan dari batuan plutonik menghasilkan batuan bertekstur **fanerik (berbutir kasar)**.



Igneous Rocks

Plutonic vs Volcanic

Klasifikasi, Texture dan Komposisi

- Klasifikasi batuan beku didasarkan pada **texture dan komposisi**.
- **Batuan beku vulkanik** bertexture afanitik atau porfiritik termasuk: **riolit, andesit, basalt dan tuff**.
- **Batuan beku plutonik** bertexture fanerik termasuk: **granit, gabbro dan diorit**.



Igneous Rocks

Komposisi dan Texture Batuan Beku

- Berdasarkan komposisi batuan beku dibagi menjadi dua yaitu: **Felsic (asam)** seperti **granit dan riolit** dan **Mafik (basa)** seperti **gabro, diorit dan basalt**.
- Sedangkan batuan beku yang komposisinya **intermedier termasuk granodiorit dan andesit**.
- **Pluton** adalah tubuh batuan beku yang mengintrusi batuan samping atau yang membeku jauh dibawah permukaan bumi. Intrusi diklasifikasikan berdasarkan volume dan geometri. Intrusi yang **konkordan termasuk sill dan lakolit**. Intrusi **diskordan termasuk dike, vulkanik neck, batolit dan stock**.
- **Batolit** adalah batuan plutonik yang besar dengan luas permukaan $\pm 100\text{km}^2$. **Stock** adalah batolit dengan ukuran jauh lebih kecil.



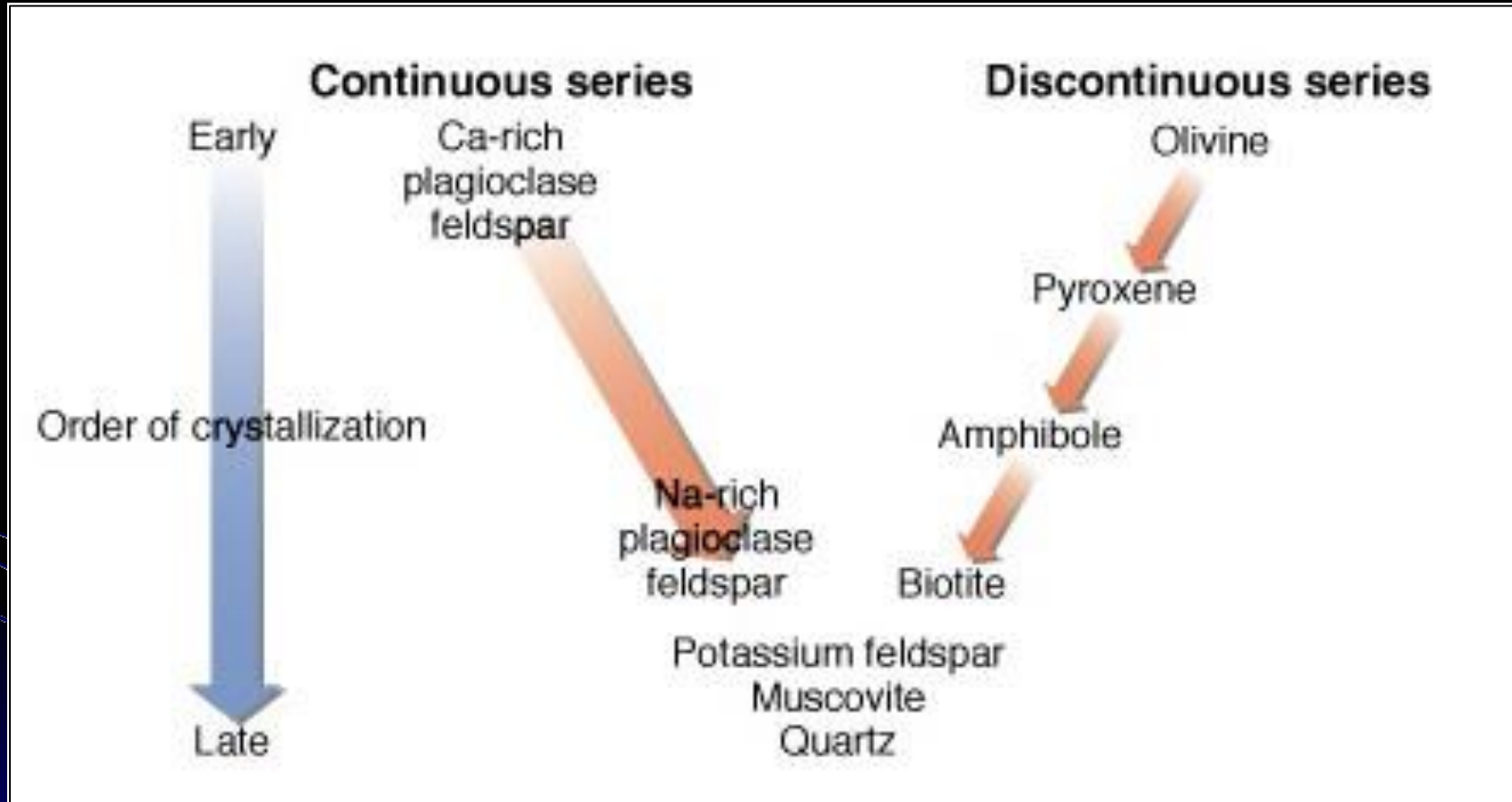
Igneous Rocks

Mafic vs Felsic

- Secara umum ada dua jenis magma :
 - **Asam (Felsic)**
 - **Basa (Mafic)**
- Pada kondisi ideal pendinginan magma yang mafic (basa) akan memberikan sekuen mineral yang berbeda yang stabil pada kondisi temperatur tertentu yang dinamakan sebagai deret atau seri **reaksi Bowen**.

Igneous Rocks

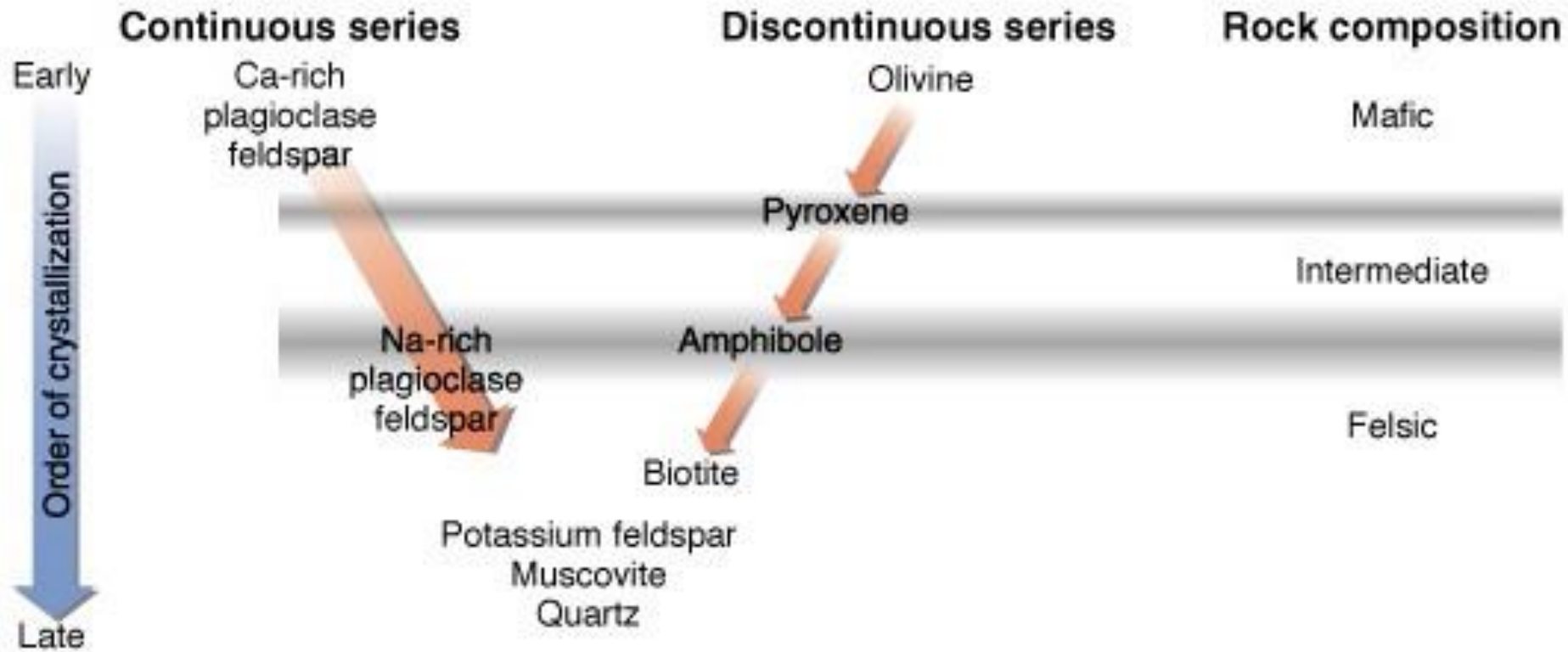
Bowen Series



Bowen Reaction Series

Igneous Rocks

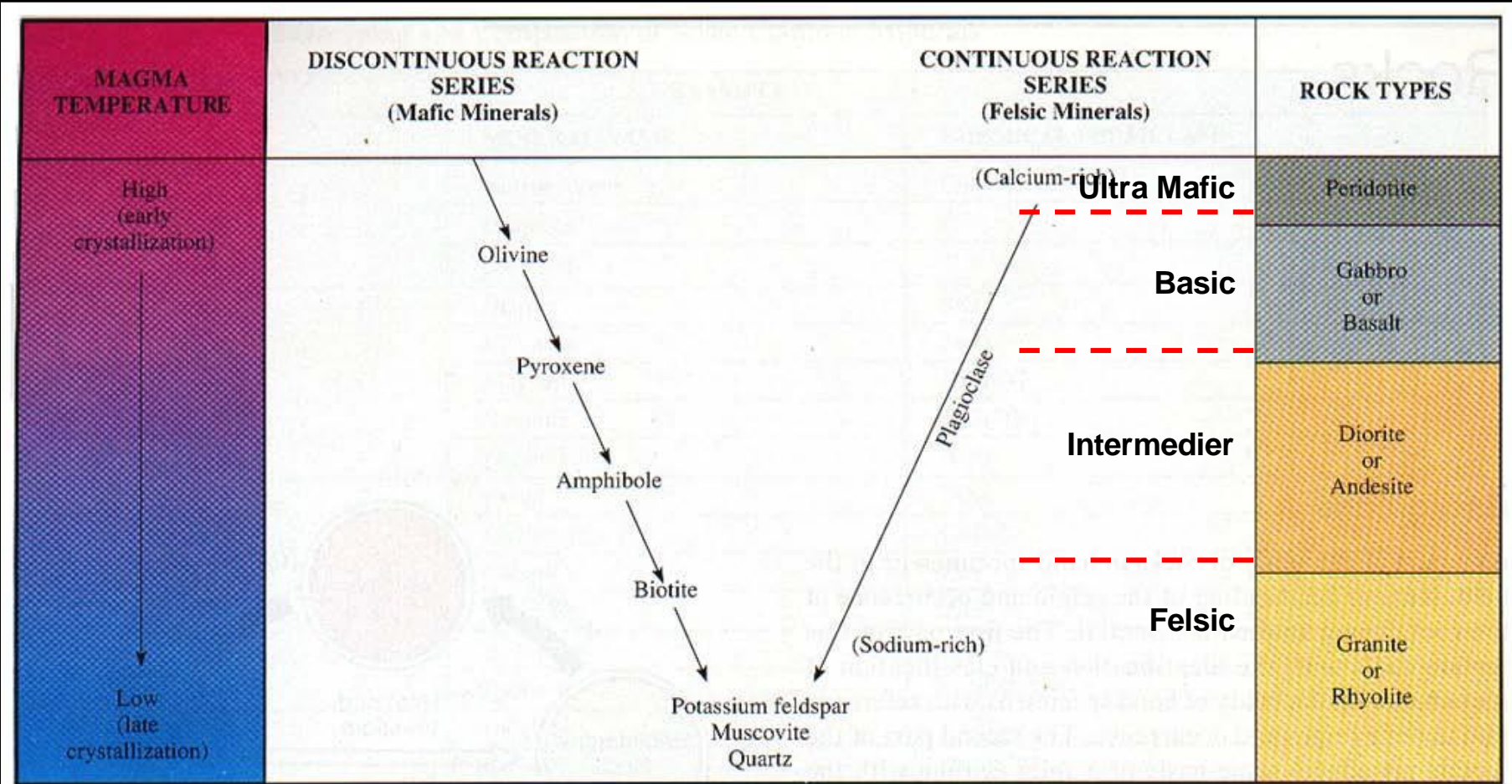
Bowen Series



Bowen Reaction Series & Major Igneous Rocks

Igneous Rocks

Bowen Series

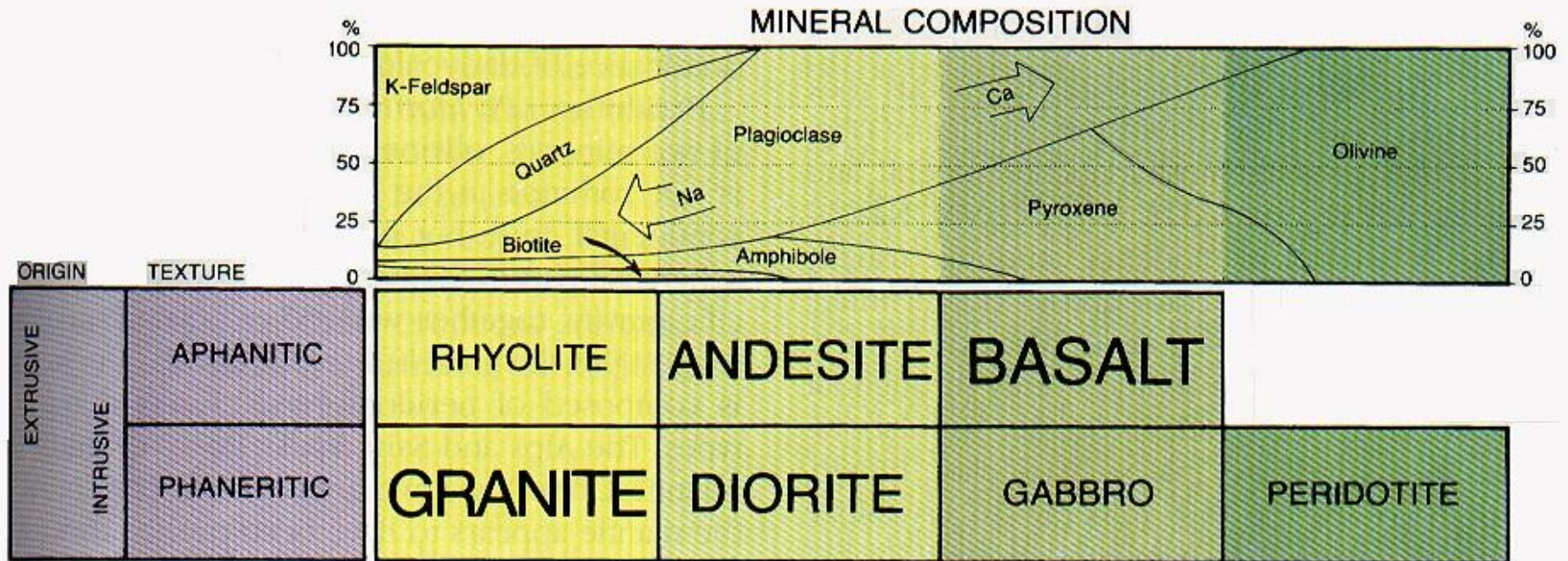


Reaction series for igneous rock formation from a magma.

Major Igneous Rocks

Igneous Rocks

Major Classification



The classification of igneous rocks is based on texture (shown vertically on the chart) and composition (shown horizontally). The size of type in which the names of the rocks are printed is roughly proportional to their abundance.

Igneous Rock Classification

Igneous Rocks

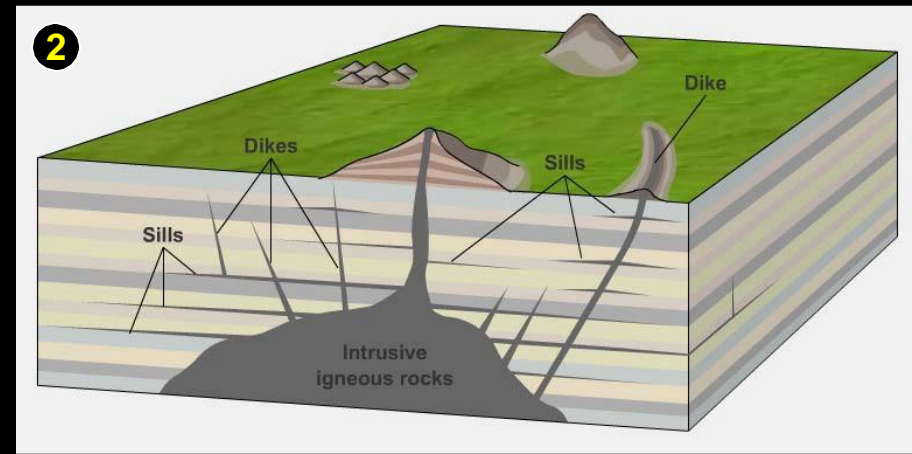
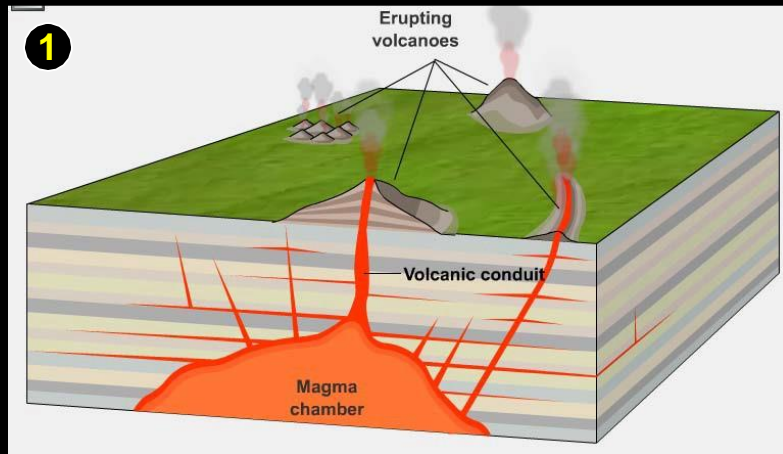
Intrusive vs. Extrusive

Relationship of Igneous Rock Types to
Their Modes of Occurrence in the Earth's Crust.

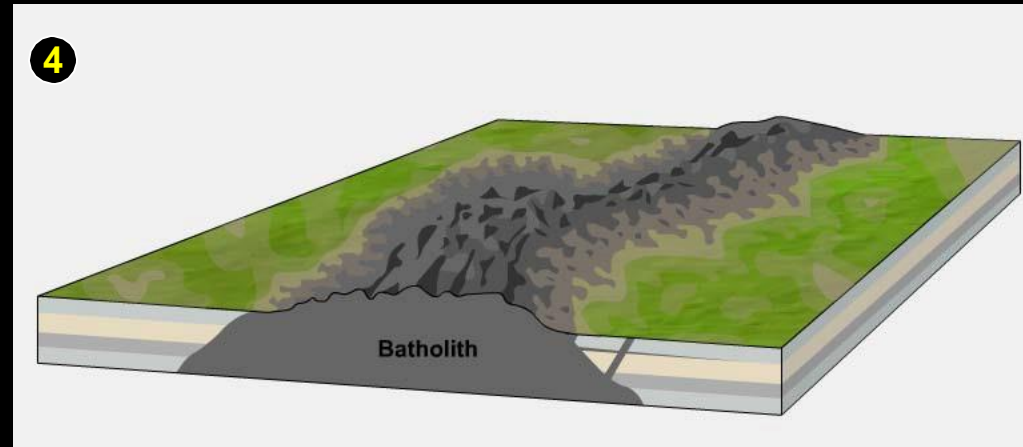
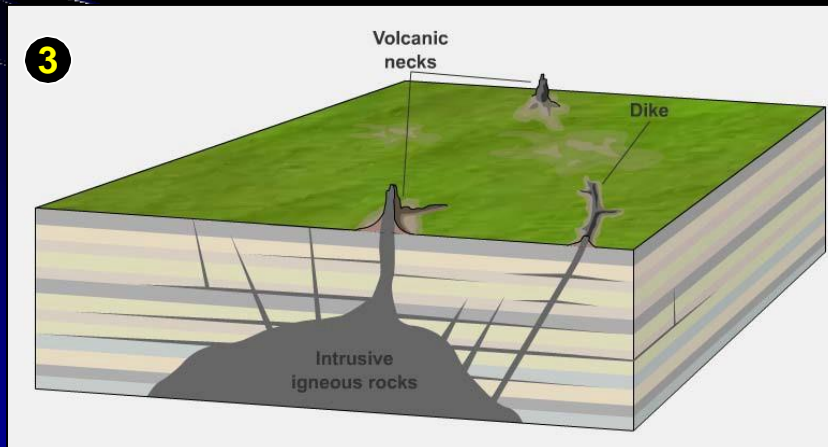
	ROCK TYPE	SOME MODES OF OCCURRENCE
EXTRUSIVE	Pumice Scoria	Lava flows, pyroclastics Crusts on lava flows, pyroclastics
	Obsidian	Lava flows
	Rhyolite Andesite Basalt	Lava flows, shallow intrusives
INTRUSIVE	Rhyolite porphyry Andesite porphyry Basalt porphyry	Dikes, sills, laccoliths, intruded at medium to shallow depths
	Granite Diorite Gabbro Peridotite	Batholiths and stocks of deep-seated intrusive origin

Igneous Rocks

Igneous Body



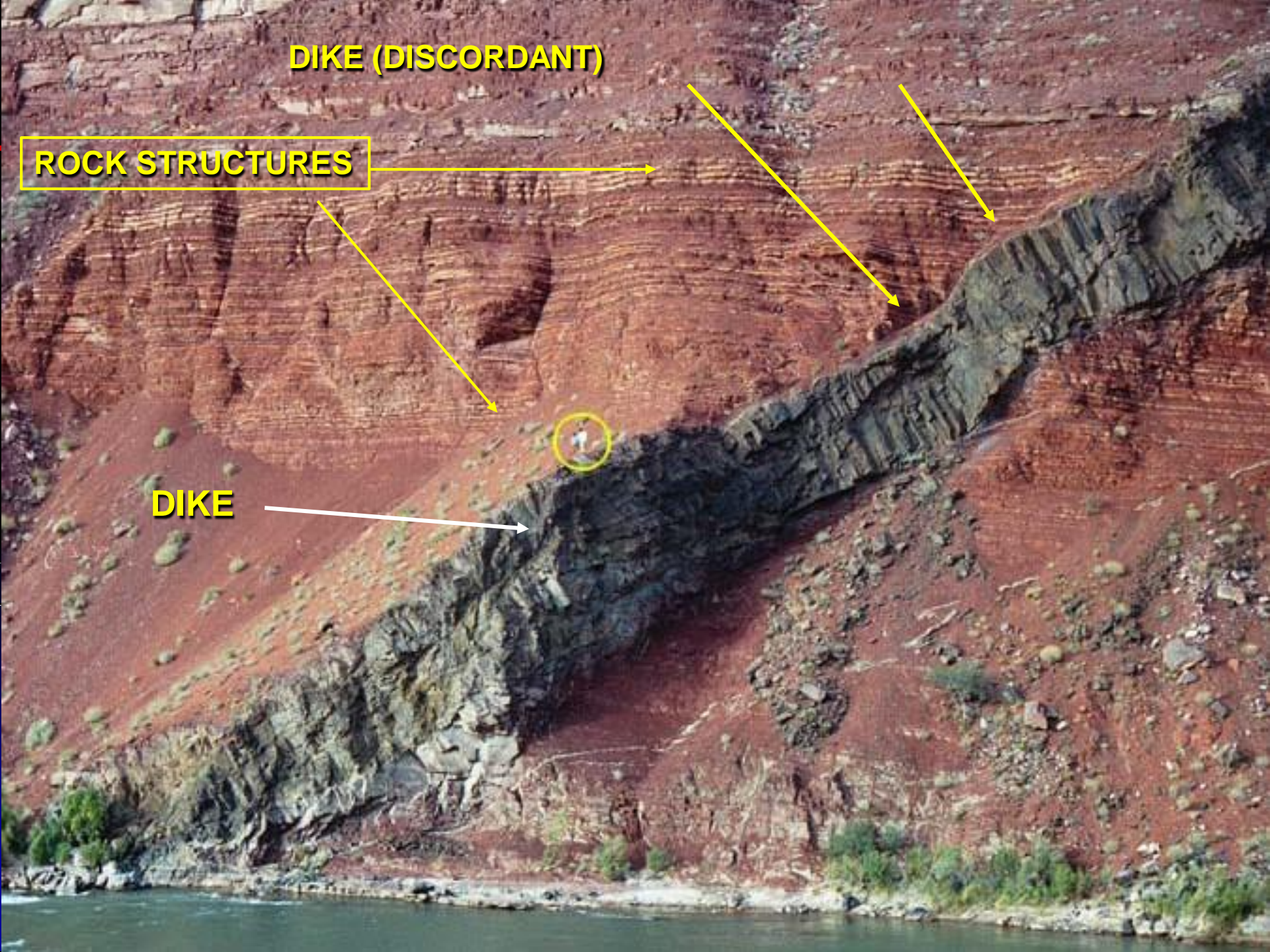
GEOMETRY OF IGNEOUS BODY



DIKE (DISCORDANT)

ROCK STRUCTURES

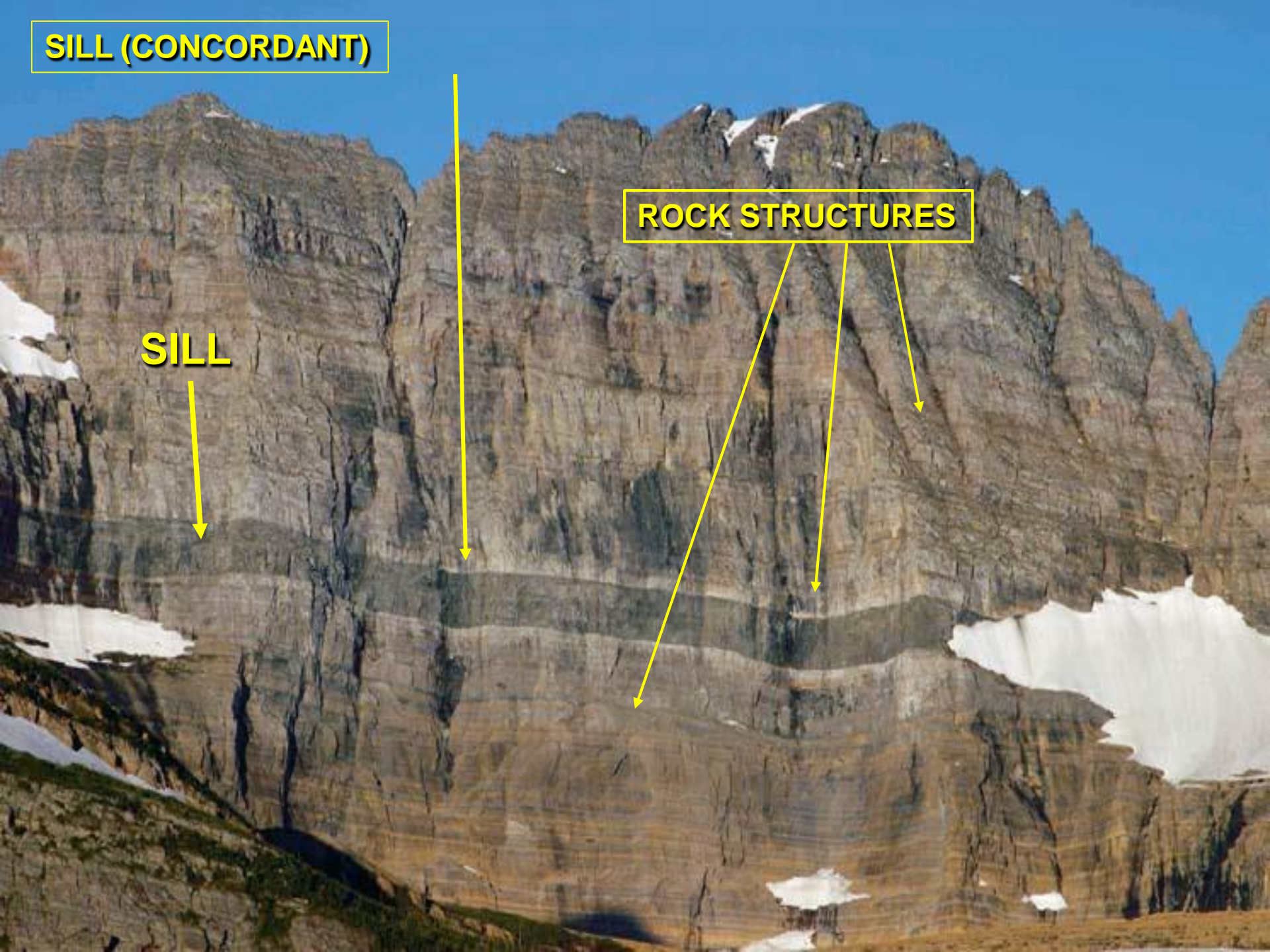
DIKE



SILL (CONCORDANT)

ROCK STRUCTURES

SILL

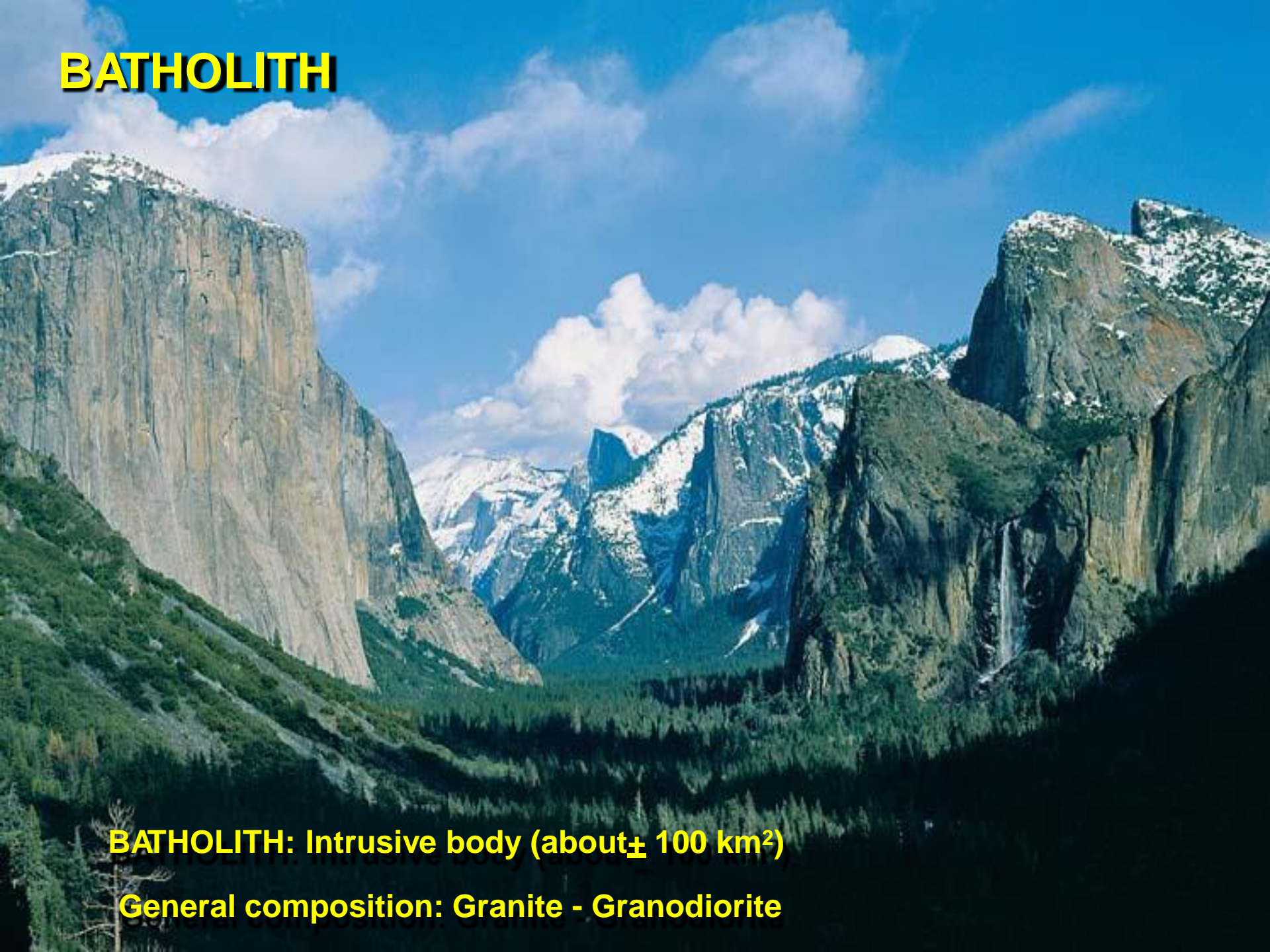


VOLCANIC NECK (PLUG)



Volcanic Neck: Volcanic plug/crater remnant in the surface

BATHOLITH

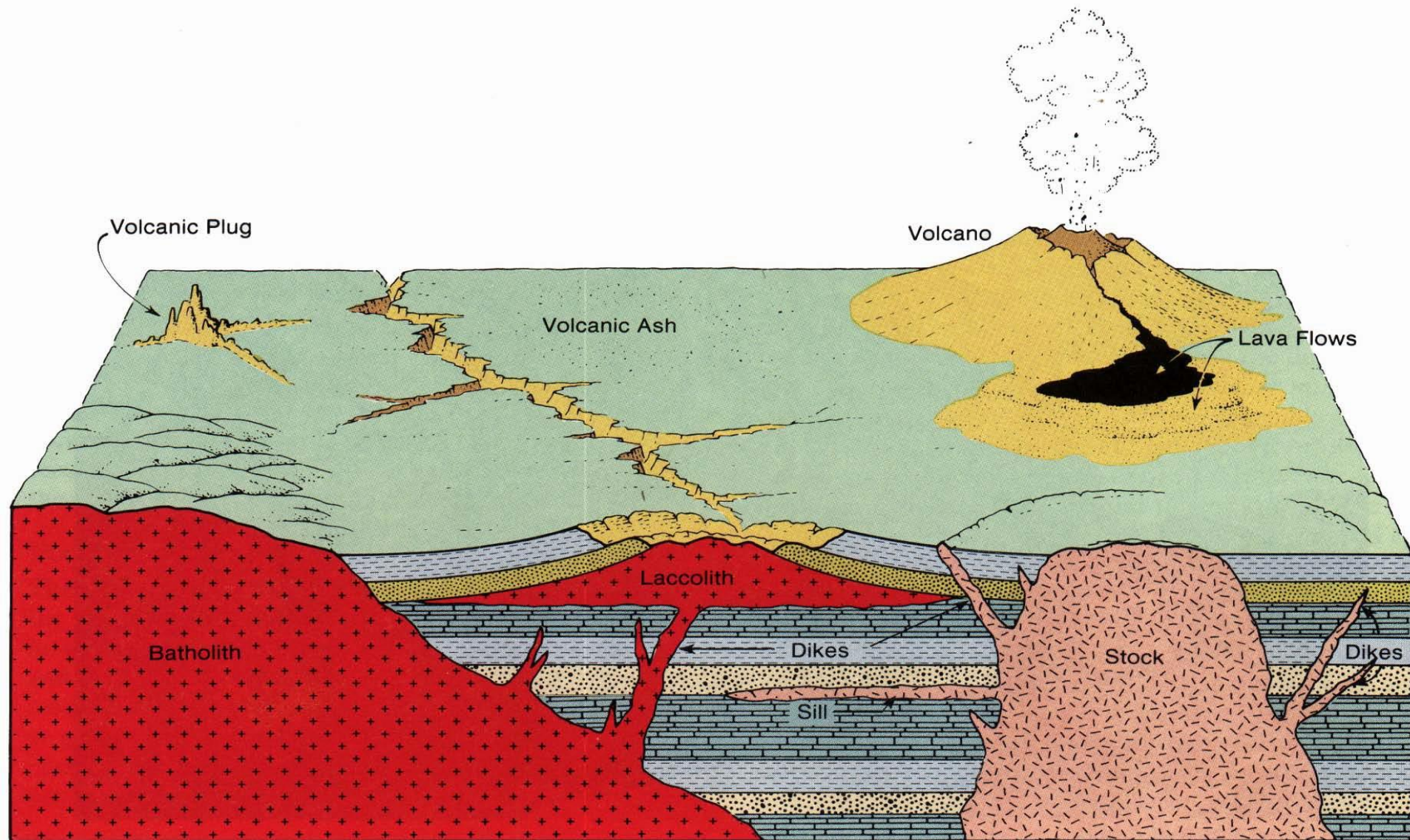


BATHOLITH: Intrusive body (about $\pm 100 \text{ km}^2$)

General composition: Granite - Granodiorite

Igneous Rocks

Igneous Body



Block diagram showing various modes of occurrence of igneous rocks.